



Fungal Origins of the Bicyclo[2.2.2]diazaoctane Ring System of Prenylated Indole Alkaloids

Finefield, Jennifer M.; Frisvad, Jens C.; Sherman, David H.; Williams, Robert M.

Published in:
Journal of Natural Products

Link to article, DOI:
[10.1021/np200954v](https://doi.org/10.1021/np200954v)

Publication date:
2012

Document Version
Early version, also known as pre-print

[Link back to DTU Orbit](#)

Citation (APA):
Finefield, J. M., Frisvad, J. C., Sherman, D. H., & Williams, R. M. (2012). Fungal Origins of the Bicyclo[2.2.2]diazaoctane Ring System of Prenylated Indole Alkaloids. *Journal of Natural Products*, 75(4), 812-833. <https://doi.org/10.1021/np200954v>

General rights

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

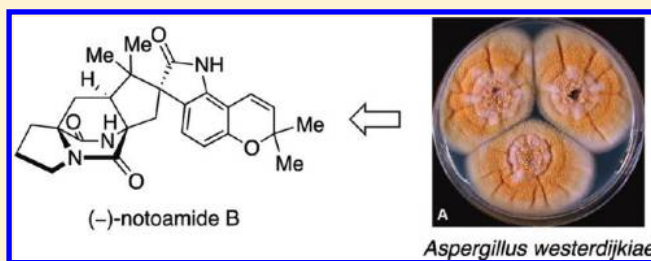
- Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the public portal

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Fungal Origins of the Bicyclo[2.2.2]diazaoctane Ring System of Prenylated Indole Alkaloids

Jennifer M. Finefield,[†] Jens C. Frisvad,[‡] David H. Sherman,[§] and Robert M. Williams^{*,†,⊥}[†]Department of Chemistry, Colorado State University, Fort Collins, Colorado 80523, United States[‡]Department of Systems Biology, Building 221, Technical University of Denmark, DK-2800 Kgs. Lyngby, Denmark[§]Life Sciences Institute and Departments of Medicinal Chemistry, Microbiology & Immunology, and Chemistry, University of Michigan, Ann Arbor, Michigan 48109, United States[⊥]University of Colorado Cancer Center, Aurora, Colorado 80045, United States

ABSTRACT: Over eight different families of natural products consisting of nearly 70 secondary metabolites that contain the bicyclo[2.2.2]diazaoctane ring system have been isolated from various *Aspergillus*, *Penicillium*, and *Malbranchea* species. Since 1968, these secondary metabolites have been the focus of numerous biogenetic, synthetic, taxonomic, and biological studies and, as such, have made a lasting impact across multiple scientific disciplines. This review covers the isolation, biosynthesis, and biological activity of these unique secondary metabolites containing the bridging bicyclo[2.2.2]diazaoctane ring system. Furthermore, the diverse fungal origin of these natural products is closely examined and, in many cases, updated to reflect the currently accepted fungal taxonomy.



■ INTRODUCTION

Natural products derived from fungal sources have been an area of intense research for decades due to the vast structural diversity and biological activity of these secondary metabolites. Numerous fungal-derived natural products serve as clinical therapeutics for both human and animal diseases, and thus the search for new classes of bioactive secondary metabolites from terrestrial and marine fungal sources has garnered additional interest.¹ Natural products containing a unique bicyclo[2.2.2]-diazaoctane ring system are part of a diverse class of biologically active prenylated indole alkaloids isolated mainly from various *Penicillium* and *Aspergillus* species of fungi.² The brevianamides,³ paraherquamide,⁴ stephacidins,⁵ notoamides,⁶ asperparalines,⁷ marcfortines,⁸ chrysogenamides,⁹ and malbrancheamides¹⁰ exhibit a range of interesting structural features, such as a bicyclo[2.2.2]diazaoctane nucleus and a complex, oxidized amino acid skeleton (Figure 1). Structurally, these natural products are comprised of tryptophan, a cyclic amino acid (proline, pipecolic acid, or derivatives of proline or pipecolic acid), and one or two isoprene units. In addition to their structural diversity, a variety of bioactivities are also present for these compounds, including but not limited to insecticidal, cytotoxic, anthelmintic, and antibacterial properties.

Initial characterization of these types of natural products led to the conclusion that two isoprene units are present in these alkaloid families, in which one of the two isoprene units is involved in the formation of the bicyclo[2.2.2]diazaoctane ring system.¹¹ In order to determine the biosynthetic pathway from which the isoprene units originate, [¹³C₂]-acetate has been utilized in precursor incorporation studies with fungal cultures

known to produce metabolites containing the bridging bicycle. Specific incorporation of intact C₂ units, combined with the ¹³C coupling patterns observed by ¹³CNMR spectroscopy, demonstrated that the isoprene units found in these metabolites arise from dimethylallyl pyrophosphate (DMAPP), a product of the mevalonic acid pathway (Scheme 1).^{11,12}

Of the natural products containing the unusual bridging bicycle, two distinct stereochemical configurations have been noted with respect to the relative configuration at the C-19 position (brevianamide numbering). The rare *anti*-configuration (1) and the more common *syn*-configuration (2) are determined by the position of the C–H proton at the bridgehead of the bicyclic core in relation to the bridging secondary lactam (Figure 2).¹³ The brevianamides are currently the only family of secondary metabolites to display the *anti*-configuration about the bridging bicycle; however, it was recently discovered that two novel metabolites chrysogenamide A and versicolamide B also display the rare *anti*-stereochemistry.^{6f,9} Contrary to this, all other known metabolites containing the core bridging bicycle display the *syn*-configuration.

Since the first isolation of these bicyclo[2.2.2]diazaoctane-containing metabolites, substantial effort has focused on both the synthetic and the biosynthetic aspects of this unique ring system.^{14,15} However, although relatively little is known about the microbial assembly and modification of metabolites containing this core bridging bicycle, new genomic and biochemical approaches are beginning to clarify important aspects

Received: December 8, 2011

Published: April 15, 2012

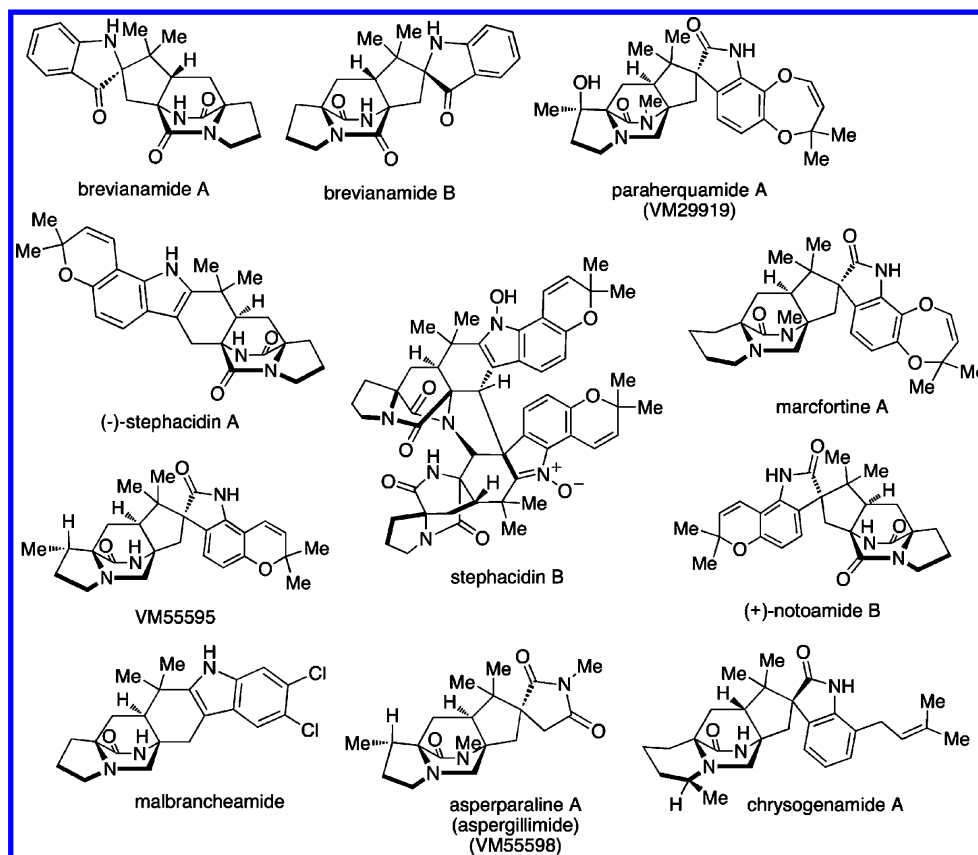
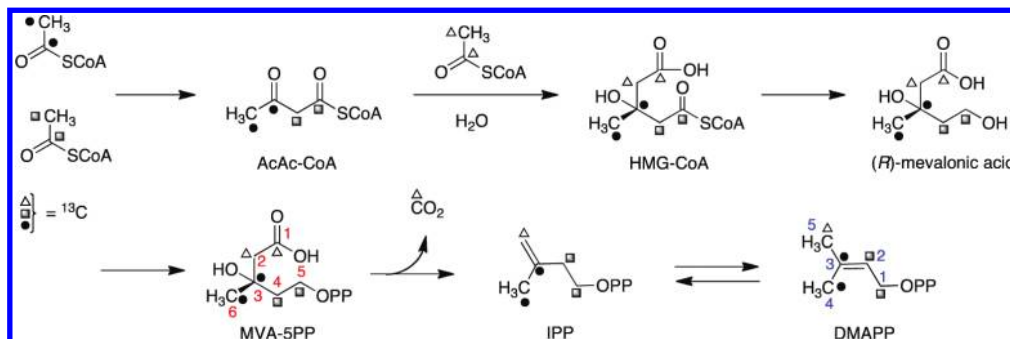


Figure 1. Structures of several secondary metabolites that contain the core bicyclo[2.2.2]diazaoctane ring system.

Scheme 1. Classical Mevalonic Acid Pathway Showing the Labeling Pattern via 1,2-Doubly Labeled Acetate^a



^aNote the change in carbon numbering.

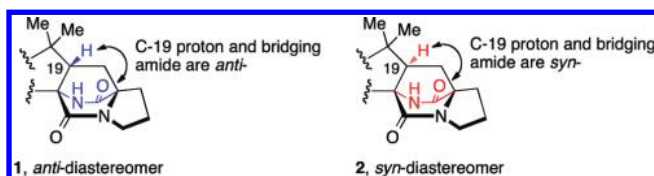
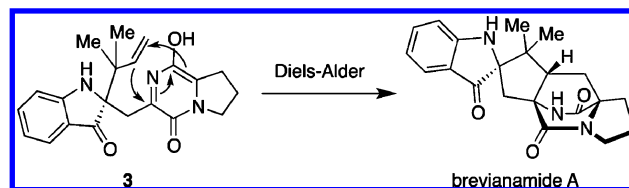


Figure 2. Anti- and syn-relationship of the bicyclo[2.2.2]diazaoctane core.

of these pathways. Originally proposed by Porter and Sammes in 1970, the biosynthetic formation of the bicyclo[2.2.2]diazaoctane ring system is believed to arise via an intramolecular hetero Diels–Alder reaction (IMDA) of a 5-hydroxypyrazine-2(1*H*)-one (Scheme 2).¹⁶ Extensive synthetic research carried out by Williams and co-workers has aimed at supporting a biosynthetic Diels–Alder reaction via the

Scheme 2. Proposed Biosynthetic Diels–Alder Reaction¹⁶



biomimetic syntheses of several advanced metabolites.^{15–17} Concurrently, substantial effort has been put forth to isolate and characterize the putative Diels–Alderase enzyme responsible for this biosynthetic transformation, but, to date, no such enzyme has been isolated.

The *Penicillium* and *Aspergillus* fungal producers of secondary metabolites containing the bicyclo[2.2.2]diazaoctane ring have been defined taxonomically based on early phenotypic methods

Table 1. Culture Collection Strains Producing Secondary Metabolites with a Bicyclo[2.2.2]diazaoctane Unit and Related Compounds (new data: Frisvad, unpublished)

secondary metabolite	fungal species	producers: isolates available in international culture collections
malbrancheamides	<i>Malbranchea aurantiaca</i> ^a <i>M. graminicola</i>	(RRC1813) ^b (UCSC 086937A)
chrysogenamide	<i>P. chrysogenum</i>	(No. 005) ^b
brevianamides	<i>P. viridicatum</i> <i>P. brevicompactum</i>	IMI 039755ii, ^c IMI 351305, IMI 351306, NRRL 869, NRRL 870, NRRL 959, NRRL 962, NRRL 3586, NRRL 3600, NRRL A-15402, NRRL A-18563, NRRL A-26909 CBS 257.29, ^c CBS 256.31, CBS 316.59, CBS 317.59, CBS 480.84, CBS 100068, CBS 110069, NRRL 867, NRRL 32579
	<i>A. carlsbadensis</i>	CBS 123894 ^c
deoxybrevianamides	<i>A. pseudoustus</i> <i>P. italicum</i> <i>P. ulaiense</i>	NRRL 5856, ^c MRC 096 ATCC 48952, CBS 339.48, CBS 278.58, CBS 349.75, CBS 489.84, CBS 136.41, CBS 210.92, CBS 722.92, CBS 262.94
paraherquamides	<i>P. brasilianum</i> <i>P. cluniae</i> <i>P. glaucoroseum</i> <i>P. cf. canescens</i>	CBS 253.55, ^c CBS 338.59 CBS 276.89 ^c NRRL 908 ^c ATCC 20841, IMI 332995
paraherquamide E	<i>A. aculeatinus</i> <i>A. fijiensis</i>	CBS 204480 CBS 313.89, ^c IMI 337664 ^a
asperparalines	<i>A. aculeatinus</i> <i>A. fijiensis</i>	ATCC 204480 = JV-23, CBS 101.43, CBS 186.67, CBS 610.78, CBS 621.78, CBS 116.80, CBS 121060, CBS 121875 CBS 313.89, ^c CBS 119.49, IMI 337664 ^a
notoamides (including (–)-versicolamide)	<i>A. bridgeri</i> <i>A. melleus</i> <i>A. ochraceus</i> <i>A. ostianus</i> <i>A. persii</i> <i>A. petrakii</i> <i>A. sclerotiorum</i> <i>A. westerdijkiae</i>	CBS 350.81 ^c CBS 546.65, ^c NRRL 394 CBS 108.08, ^c CBS 589.68, CBS 624.78, NRRL 6319, (CL41582) CBS 103.07, ^c CBS 101.23, CBS 311.80, IMI 177964b, IMI 348937, NRRL 422, NRRL 5225, NRRL A-15156, (01F313) CBS 112795 ^c CBS 105.57 ^c CBS 549.65, ^c ATCC 18413, NRRL 5167, (CNC358) CBS 112803 ^c = NRRL 3174, CBS 112791, CBS 112804, NRRL 5175, NRRL 5221, NRRL 5228,
notoamides (including (+)-versicolamide)	<i>A. protuberus</i> <i>A. versicolor</i> <i>A. elongatus</i>	CBS 602.74, ^c CBS 601.74 CBS 118.50, CBS 106.57, CBS 599.65, IMI 096225, NRRL 35600 CBS 387.75 ^c
austamide	<i>A. pseudoustus</i>	NRRL 5856, ^c MRC 096
citrinadins	<i>P. citrinum</i>	CBS 139.45, ^c (N-059)
PF1270 citrinadins	<i>P. westlingii</i>	(PF1270)
talathermophilins	<i>Talaromyces thermophilus</i>	CBS 116.72, (YM 1–3)

^aPossible misidentification. ^bStrain unavailable. ^cType strain.

of Raper and Thom¹⁸ for *Penicillium* and Raper and Fennell¹⁹ for *Aspergillus*. However, several of these species have since been reclassified based on modern molecular methods. Extensive taxonomical research carried out by Frisvad and co-workers has provided new phylogenetic information for each fungal culture known to produce the bicyclo[2.2.2]diazaoctane ring system described in this review, and both the original and updated fungal strain nomenclature will be mentioned when applicable. Furthermore, numerous strains of these fungal cultures have been determined to produce the same sets of secondary metabolites (Table 1). Taxonomic revisions of species and reidentifications of the strains are based on a polyphasic approach, including morphological, physiological, chemical, and DNA sequence data. Mostly sequences from β -tubulin and calmodulin genes have been valuable in assigning strains to new or known species.

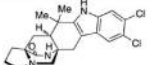
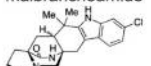
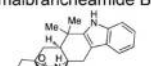
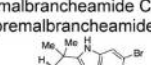
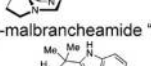
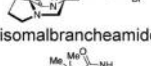
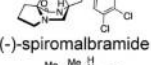
■ MALBRANCHEAMIDES

First identified in 2006 by Mata and co-workers, malbrancheamide and malbrancheamide B were isolated from *Malbranchea aurantiaca* RRC1813, a fungus obtained from bat guano found

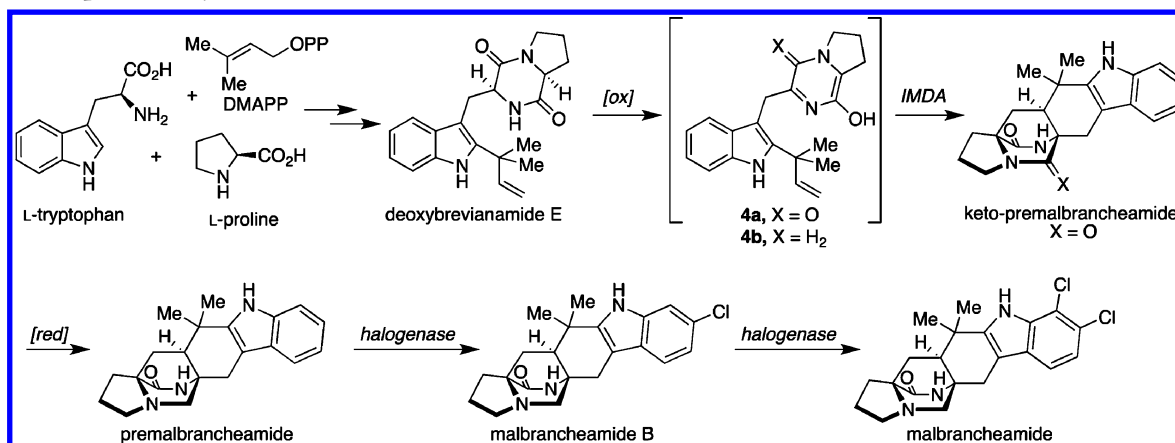
in a cave in Mexico (Table 2).¹⁰ Shortly following the isolation of malbrancheamide and malbrancheamide B, Mata and co-workers reported the presence of the nonhalogenated precursor malbrancheamide C, which was later named premalbrancheamide.^{10c} This unique family of halogenated secondary metabolites has expanded recently with the isolation of two new chlorinated metabolites, (–)-spiromalbramide and (+)-isomalbrancheamide B, from *Malbranchea graminicola*, as reported by Crews and co-workers.^{10d} Furthermore, two brominated analogues, (+)-malbrancheamide C (hereafter referred to as (+)-malbrancheamide “C”) and (+)-isomalbrancheamide C, were obtained from *M. graminicola* when the growth medium was enriched with bromine salts.

This family of natural products represents the only bicyclo[2.2.2]diazaoctane-containing metabolites to be reported from a fungal source outside the genera *Penicillium* and *Aspergillus*. Taxonomically, the strain of *M. aurantiaca* (RRC1813) may either be that species or represent the species *M. graminicola*, since three strains of *M. aurantiaca* from the CBS collection (CBS 127.77, Utah, ex culture contaminant, type strain; CBS 759.74 and CBS 309.76, both ex soil, India) did not produce

Table 2. Fungal Origin of the Malbrancheamides

compound	fungal species	bioactivity	reference
 malbrancheamide	<i>M. aurantiaca</i> RRC1813	calmodulin inhibitor	10a, 20
 malbrancheamide B	<i>M. aurantiaca</i> RRC1813	weak calmodulin inhibitor	10b, c, 20
 malbrancheamide C (premalbrancheamide)	<i>M. aurantiaca</i> RRC1813		10c, f
 (+)-malbrancheamide "C"	<i>M. graminicola</i> UCSC 086937A		10d
 (+)-isomalbrancheamide C	<i>M. graminicola</i> UCSC 086937A		10d
 (-)-spiromalbramide	<i>M. graminicola</i> UCSC 086937A		10d
 (+)-isomalbrancheamide B	<i>M. graminicola</i> UCSC 086937A		10d

Scheme 3. Proposed Biosynthesis of the Malbrancheamides



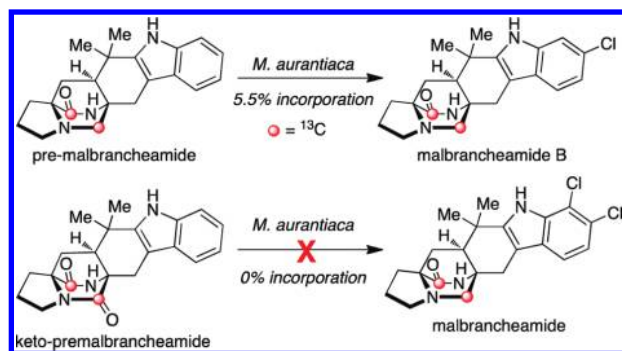
the malbrancheamides or penicillic acid.^{10c} However, the isolation of (+)-malbrancheamide, (+)-malbrancheamide B, (+)-premalbrancheamide, and (–)-spiromalbramide from *M. graminicola* indicates that these compounds are common in the genus *Malbranchea*.

Structurally, the malbrancheamides are the first members of this class of metabolites to display halogenation of the indole moiety. While malbrancheamide was determined to contain chlorination at the C-5 and C-6 position of the indole, the exact structure of malbrancheamide B was unclear. By synthesizing both the C-5-chlorinated and C-6-chlorinated potential compounds, it was established that malbrancheamide B contains C-6 monochlorination.²¹ In conjunction with the unique indole chlorination, the malbrancheamides are also characterized by the *syn*-configuration about the bridging bicycle, as well as the presence of a monoketopiperazine moiety in the core bicycle. Biologically, the malbrancheamides were determined to display calmodulin inhibitory activity of CaM-dependent phosphodiesterase (PDE1).^{10c,f}

Following the isolation of the malbrancheamides, interest was sparked in elucidating the biosynthetic timing of the indole chlorination. As shown in Scheme 3, the initially proposed biogenesis of the malbrancheamides begins with the condensation of tryptophan, proline, and dimethylallyl pyrophosphate to afford the known natural product deoxybrevianamide E. From here, two distinct biosynthetic pathways are possible to form the known metabolite premalbrancheamide. One potential pathway involves oxidation of deoxybrevianamide E to yield the Diels–Alder precursor 5-hydroxypyrazin-2(1H)-one (**4a**), which would undergo [4+2] cycloaddition to afford keto-premalbrancheamide. Subsequent carbonyl reduction of the tertiary amide would provide premalbrancheamide. On the other hand, carbonyl reduction could occur prior to the IMDA reaction, in which **4b** would serve as the key achiral intermediate and direct intermediate to premalbrancheamide. Two individual and concurrent halogenases would then lead from premalbrancheamide to malbrancheamide B and then to malbrancheamide.

To test whether premalbrancheamide and/or keto-premalbrancheamide serve as biosynthetic precursors to the halogenated metabolites, doubly ^{13}C -labeled premalbrancheamide and keto-premalbrancheamide were synthesized by Williams et al. and provided to *M. aurantiaca* in separate precursor incorporation studies (Scheme 4).^{10f} Results from

Scheme 4. Precursor Incorporation Studies with *M. aurantiaca*.



these feeding studies showed significant incorporation of $^{13}\text{C}_2$ -premalbrancheamide into malbrancheamide B (5.5% incorporation), but surprisingly, $^{13}\text{C}_2$ -keto-premalbrancheamide did not incorporate. The latter result suggests that the timing for the reduction of the tryptophan-derived amide carbonyl residue must occur prior to the [4+2] cycloaddition; thus the biosynthetic formation of premalbrancheamide most likely arises from the proposed intermediate **4b**. Future genetic and biochemical analysis of the malbrancheamide biosynthetic pathway will undoubtedly resolve the mechanism of monoketopiperazine assembly by *M. aurantiaca*.¹

■ CHRYSOGENAMIDE

In 2008, Zhu and co-workers were interested in identifying new neuroprotective compounds from endophytic fungi on the parasitic plant *Cistanche deserticola* collected in northwest mainland China (Table 3).⁹ From the roots of *C. deserticola*, three active strains were screened for neuroprotective effects on SH-SY5Y cells, and the fungal culture *Penicillium chrysogenum* No. 005 demonstrated significant activity. From *P. chrysogenum*, the novel metabolite chrysogenamide A was isolated. From a taxonomic perspective, even though the original culture was not available for examination, other secondary metabolites characteristic of *P. chrysogenum* were also isolated,⁹ supporting the identity.^{22–24}

From a structural point of view, this novel metabolite is closely related to the marcfortines (discussed below) in that chrysogenamide A was found to contain a bicyclo[2.2.2]-diazaoctane ring system, a spiro-oxindole moiety, and a pipercolic acid unit with C-17 methylation. Additionally, chrysogenamide A displays the rare *anti*-configuration about the bridging bicycle and an isoprene unit at the C-7 indole position, as opposed to the more commonly observed dioxepin

and pyran rings. Chrysogenamide A is the first and only metabolite of these families of natural products to display neurocyte protection against oxidative stress-induced cell death.⁹

■ BREVIANAMIDES

Between 1969 and 1972, Birch and co-workers isolated six novel secondary metabolites, brevianamides A–F (Figure 3),

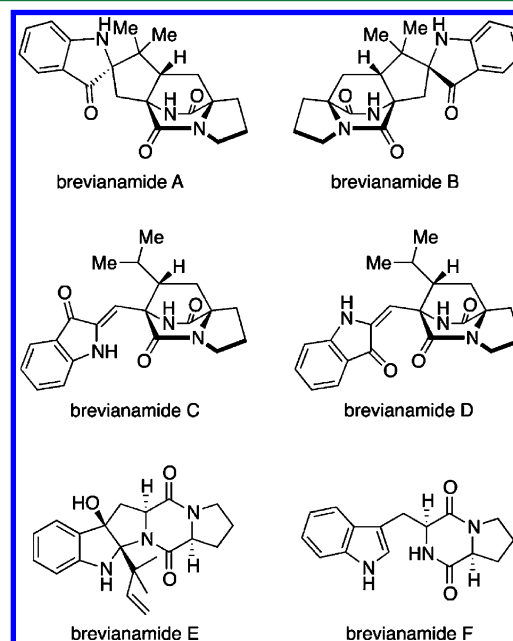


Figure 3. Structures of the brevianamides and related metabolites.

from *Penicillium brevicompactum*.³ Of these six metabolites, four compounds were found to contain the unique bicyclo[2.2.2]-diazaoctane core. Additional research revealed that brevianamides A and B are diastereoisomers, while brevianamides C and D are artifacts of isolation due to light irradiation.^{3c} Structurally, the brevianamides were found to display the *anti*-configuration about the bicyclic core, as well as the spiro-indoxyl moiety, and an unsubstituted indole ring. Furthermore, the brevianamides are a class of natural products that display modest biological activities, in which brevianamides A and D have been shown to possess antifeedant and insecticidal effects.²⁵

As shown in Table 4, the brevianamides have been isolated predominately from two *Penicillium* species. The two main producers of brevianamide A are *P. brevicompactum*^{3,22,23} and *P. viridicatum*,^{3e,22,23} of which both are consistent producers of brevianamide A.²⁶ The isolate of *P. ochraceum* reported to produce brevianamide A^{3d} was determined to be a brown-spored mutant of *P. viridicatum*.²⁶ Deoxybrevianamide E is produced by *P. italicum*,^{27d,e} *P. ulaiense*,²² and a fungus identified as *Aspergillus ustus* NRRL 5856.^{27a–c} The latter isolate has been the basis for the description of a new species, *Aspergillus pseudoustus*;²⁸ however, *A. pseudoustus* is the only species known to produce austamide (discussed below). The first biosynthetic

Table 3. Fungal Origin and Bioactivity of Chrysogenamide A

compound	fungal species	bioactivity	reference
 chrysogenamide A	<i>P. chrysogenum</i> No. 005	neuroprotective effect on SH-SY5Y cells	9

Table 4. Fungal Origin of the Brevianamides

compound	fungal species	bioactivity	reference
brevianamide A	<i>P. brevicompactum</i> , <i>P. viridicatum</i> , <i>P. ochraceum</i> ^a	antifeedant insecticidal	3a–c, e, d, 25
brevianamide B	<i>P. brevicompactum</i>		3a–c
brevianamide C	<i>P. brevicompactum</i>		3a–c
brevianamide D	<i>P. brevicompactum</i>	insecticidal	3a–c, 25
brevianamide F ^b	<i>A. versicolor</i> , <i>P. brevicompactum</i>	antibacterial	6f, 3c, 30
brevianamide E	<i>P. brevicompactum</i>		3a–c
deoxybrevianamide E	<i>Aspergillus</i> sp. MF297-2, ^c <i>A. ustus</i> , ^d <i>P. ulaiense</i> , <i>P. italicum</i>		6a, 22, 27

^aMisidentified fungal culture; correct identification = *Penicillium viridicatum*. ^bWidely distributed biosynthetic precursor; thus only relevant fungal origins are listed. ^cUndergoing reidentification (possibly *Aspergillus sclerotiorum*). ^dReclassified as *Aspergillus pseudoustus*.

precursor is the common dioxopiperazine brevianamide F (cyclo-L-Trp-L-Pro). This alkaloid has been shown to serve as the precursor for several related secondary metabolites, those lacking the bridging bicycle, such as the fumitremorgins and spirotryprostatins in *Aspergillus fumigatus*.²⁹

Following the isolation of these unique natural products in 1969, which were the first metabolites to be isolated to display the unusual bicyclo[2.2.2]diazaoctane ring system, Birch and co-workers carried out several radiolabeled precursor incorporation studies in order to elucidate early stages of the biosynthetic pathway of the brevianamides.^{3a–c,31} Birch demonstrated that the brevianamides are derived biogenetically from tryptophan, proline, and mevalonic acid and later showed that radiolabeled cyclo-L-Trp-L-Pro (brevianamide F) is also a precursor to brevianamide A in cultures of *P. brevicompactum*. On the basis of these results, Birch proposed that brevianamide F arises from the condensation of tryptophan and proline, which is followed by C-2 reverse prenylation to afford deoxybrevianamide E. Oxidation and cyclization of deoxybrevianamide E would give rise to brevianamides A and B (Scheme 5).

Williams later demonstrated through tracer incorporation studies with *P. brevicompactum* that radiolabeled deoxybrevianamide E incorporates into brevianamides A, B, and E in significant radiochemical yields.³² At the same time, Williams showed that neither tritium-labeled brevianamide E nor the proposed racemic, ¹³C-labeled cycloadduct **5** incorporates into brevianamide A or B when provided to *P. brevicompactum* in isotopically labeled precursor incorporation studies (Scheme 6). These results led to the currently proposed biosynthetic pathway of brevianamides A and B, as shown in Scheme 6. Since cycloadduct **5** does not incorporate into the brevianamides, it was suggested that stereoselective indole oxidation must occur prior to the formation of the bicyclo[2.2.2]diazaoctane ring

system. Thus, it was proposed that deoxybrevianamide E is first converted to 3R-hydroxyindolenine (**6**) via an *R*-selective indole oxidase, which could then proceed along one of three pathways. Following an irreversible nucleophilic addition, **6** could be readily converted to brevianamide E, or **6** could undergo a pinacol-type rearrangement to indoxyl **7**. A second oxidation–enolization sequence would then yield the *chiral* and optically pure azadiene (**3**), followed by an IMDA reaction to directly give brevianamides A and B. It should be noted that efforts to substantiate this alternative biosynthetic pathway remain elusive due to complications regarding the synthesis of **7**. Despite extensive synthetic efforts, **7** appears to be unstable and spontaneously suffers a retro-Michael reaction to give the dehydroalanine piperazinedione and the reverse-prenylated indoxyl.

Alternatively, reordering these steps would involve the 3R-hydroxyindolenine (**6**) undergoing two-electron oxidation and enolization to the corresponding azadiene followed by intramolecular Diels–Alder cycloaddition and final pinacol-type rearrangement of the Diels–Alder cycloadducts, each of which was constituted by a 3R-hydroxyindolenine.

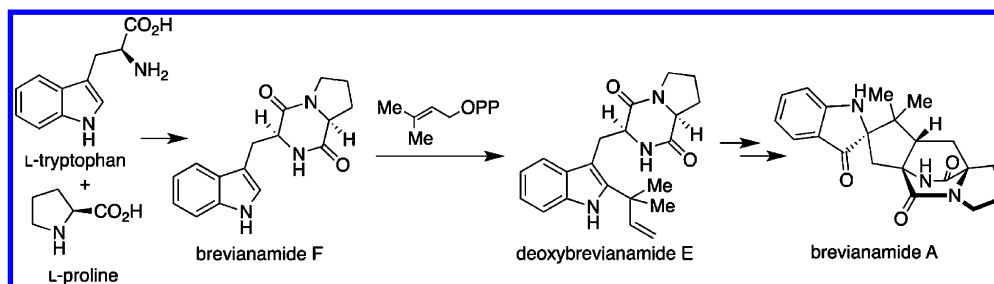
MARCFORTINES

In 1980, Polonsky and co-workers reported the isolation of three new alkaloids from *Penicillium roqueforti* (later reclassified as *P. paneum*), an essential fungus used in the production of many varieties of blue cheese that contain mold.^{8a,b} From the strain B26 of *P. paneum*, marcfortines A, B, and C were isolated (Figure 4). From a structural point of view, these metabolites contain the bicyclo[2.2.2] ring system with the *syn*-configuration and a monoketopiperazine ring. Structurally resembling chrysogenamide A, the marcfortines contain a spiro-oxindole moiety and an unsubstituted pipercolic acid unit. Furthermore, all three marcfortine natural products display substitution on the indole ring at the C-6 and C-7 positions; marcfortines A and B both contain a dioxepin ring, whereas marcfortine C has a pyran moiety.

As shown in Table 5, the marcfortines were originally isolated from a strain identified as *P. roqueforti*;^{8a,b} however, this species was subsequently subdivided into three distinct species, *P. roqueforti*, *P. carneum*, and *P. paneum*,³³ and it was determined that only the latter can produce marcfortines.^{8d,e,22,23} Additionally, *Aspergillus carneus* has been reported to produce marcfortine A;^{4c} however, the isolate producing marcfortine A and acyl aszonalenin differed from the culture ex type of *A. carneus*, which produced citrinin. Therefore, the identity of the marcfortine-producing *Aspergillus* is questionable.

The biosynthesis of marcfortine A has been investigated by Kuo et al., and, as shown in Scheme 7, this metabolite is derived from L-tryptophan (oxindole moiety), L-methionine (via SAM methylation of the monoketopiperazine N), L-lysine (pipercolate residue), and acetate (isoprene units).³⁴ On the basis of these

Scheme 5. Birch's Proposed Biosynthesis of the Brevianamides



Scheme 6. Alternative Biosynthetic Pathway for the Brevianamides

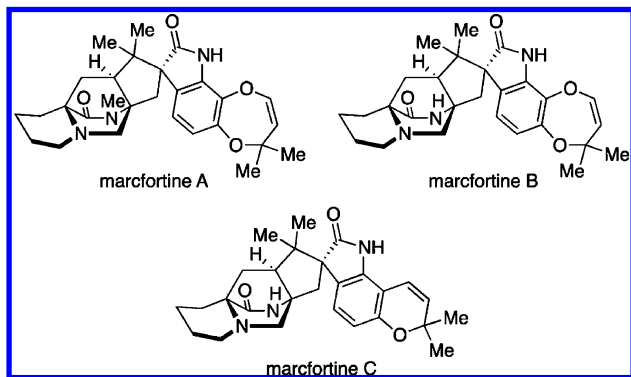
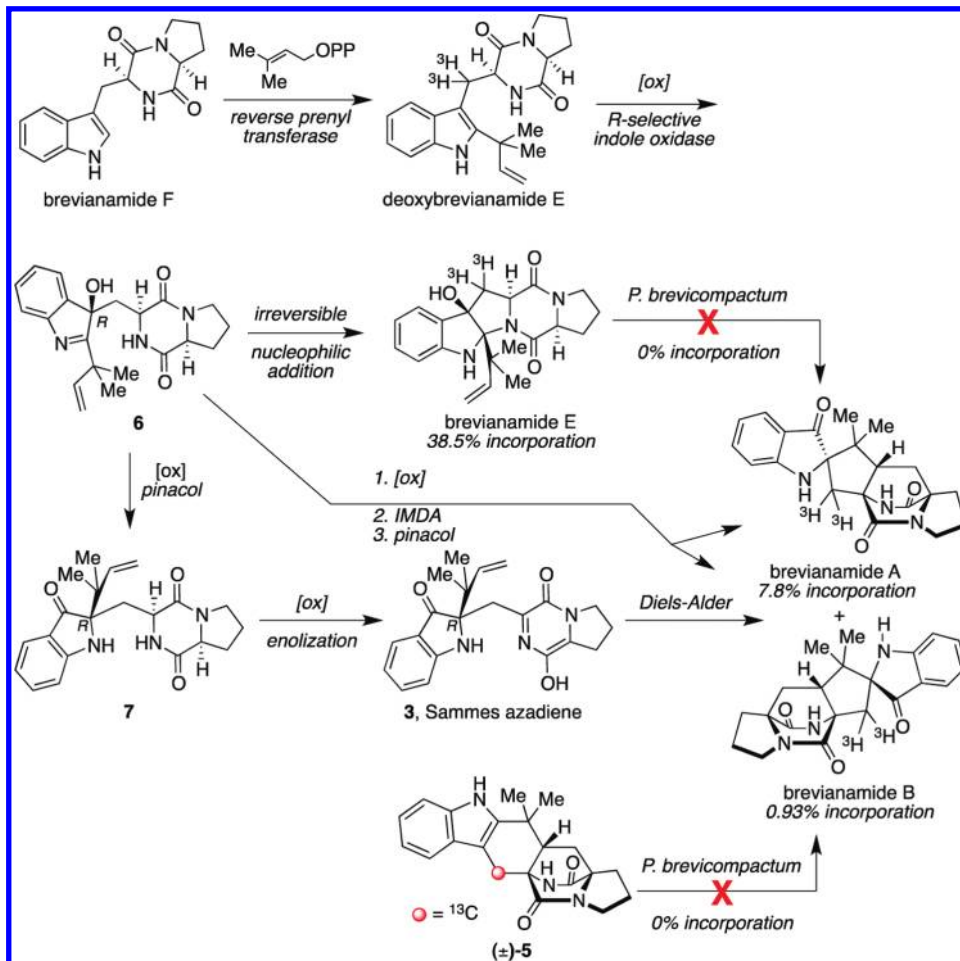


Figure 4. The marcfortine family of natural products.

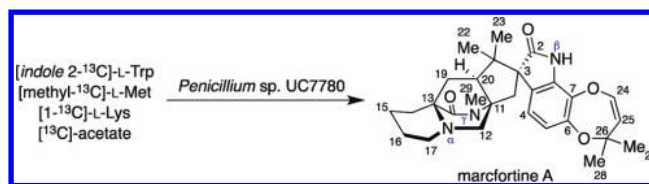
Table 5. Fungal Origin of the Marcfortines

compound	fungus species	bioactivity	references
marcfortine A	<i>A. carneus</i> , <i>P. paneum</i> , <i>P. roqueforti</i> ^a	anthelmintic	8c—e, a, b, f
marcfortine B	<i>P. paneum</i> , <i>P. roqueforti</i> ^a		8d, e, a, b
marcfortine C	<i>P. paneum</i> , <i>P. roqueforti</i> ^a		8d, e, a, b

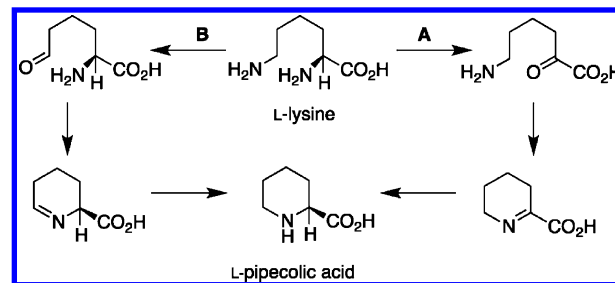
^aReclassified as *Penicillium paneum*.

results, it is clear that the isoprene residues originate from mevalonic acid; however, there are two possible biosynthetic routes for the formation of pipecolic acid from L-lysine, as shown in Scheme 8.^{34b} To determine which pathway

Scheme 7. Biosynthetic Derivatives of Marcfortine A¹⁴



Scheme 8. Two Biogenetic Routes from L-Lysine to Pipecolic Acid^{34b}



metabolizes L-lysine to the pipecolic acid unit, Kuo and co-workers carried out separate feeding studies with [α - ^{15}N]- and [ϵ - ^{15}N]-L-lysine. NMR and MS analysis revealed that the [α - ^{15}N]-L-lysine was incorporated at the α , β , and γ nitrogen of marcfortine A, with a rate of 1.8%, 13%, and 9%, respectively. Contrary to this result, [ϵ - ^{15}N]-L-lysine showed an incorporation rate of 54%, 2.7%, and 3.9% respectively, for the three

nitrogens, thus leading to the conclusion that the [ϵ - ^{15}N]-L-lysine gave a more specific and higher incorporation at the pipicolate nitrogen. On the basis of these data, pathway A is the preferred method of L-lysine metabolism to the pipicolate acid moiety.^{34b}

■ PARAHERQUAMIDES

One of the largest groups of metabolites containing the bicyclo[2.2.2]diazaoctane ring system is the paraherquamides, which were first isolated from *Penicillium paraherquei*,^{4a} a synonym of *P. brasilianum*,³⁵ and later from an isolate ATCC 20841, originally identified as *P. charlesii*.^{4h} However, *P. charlesii* was stated incorrectly³⁶ to be a synonym of *P. fellutanum* by Pitt,³⁷ and ATCC 20841 has therefore been cited as belonging to one of these species in later publications. Consequently, this isolate and another similar isolate producing paraherquamides (IMI 332995)^{4e} represent a new species closely related to *P. canescens* and will be called hereafter *P. cf. canescens*. Other producers of paraherquamides include *P. cluniae*^{4d} and *Aspergillus aculeatinus*.^{4j}

Paraherquamide A (Figure 5) was the first member of this family to be isolated by Yamazaki et al. from *P. paraherquei* (= *Penicillium brasilianum*).^{4a} A decade later, paraherquamide A was isolated from a fungus identified as *P. charlesii* (= *P. fellutanum sensu* Pitt³⁷) along with six structurally related novel analogues, paraherquamides B–G.^{4b,c} The original isolate (ATCC 20841) and a new isolate (IMI 332995) both represent a new species, provisionally called *P. cf. canescens*. In the following years, the paraherquamide family of prenylated indole alkaloids continued to expand with the isolation of paraherquamides H and I and several related compounds being reported, including VM55595, VM55596, VM55597, VM55599, SB200437, and SB203105.⁴ More recently, Sherman, Williams, and co-workers reported the occurrence of pre-paraherquamide in cultures of the paraherquamide A-producing *P. cf. canescens*, as well as in the asperparaline A-producing fungus *Aspergillus japonicus* (later identified as *A. aculeatinus*).^{4j} Furthermore, it was determined that paraherquamides A and B are also produced by *A. aculeatinus*. *P. brasilianum*, *P. cf.*

canescens, and *P. cluniae*, which are all soil fungi and the main producers of paraherquamides (Table 6).

Structurally, all of the paraherquamides contain the bicyclo[2.2.2]diazaoctane core with the *syn*-configuration; however, the paraherquamide family displays a plethora of skeletal diversity. For example, the paraherquamides exhibit varying substitution and oxygenation on not only the proline ring but also the spiro-oxindole moiety, in that several of the paraherquamides contain an unusual dioxepin ring, while others contain a pyran moiety. Within this family of secondary metabolites, VM55599 and pre-paraherquamide are the only two compounds that do not contain either a spiro-oxindole functionality or an oxygenated indole ring system.

The paraherquamides have attracted considerable interest due to their potent anthelmintic activity. Parasitic nematodes cause severe health problems in humans and domestic animals, and the severity of this problem is increasing due to the development of drug-resistant strains of parasites.³⁸ Initial anthelmintic activity was detected in *Trichostrongylus colubrifomis*,³⁹ and subsequent investigations showed that paraherquamide A is effective against strains of parasites that are resistant to broad-spectrum anthelmintics.⁴⁰ However, the toxicity of paraherquamide A was determined to be too great for use as a broad-spectrum anthelmintic.⁴¹ To decrease the toxicity, but still maintain potency, several paraherquamide derivatives have been synthesized and tested for anthelmintic activity.³⁸ Of these derivatives, 2-deoxyparaherquamide A (Figure 6), later named derquantel (Pfizer), displayed excellent activity in models of *Hemonchus contortus*, *T. colubriformis*, and *Ostertagia ostertagi*, while maintaining a safe level of toxicity. Manufacturing of 2-deoxyparaherquamide relies on a semi-synthetic procedure wherein paraherquamide is isolated from large-scale fermentation of *P. simplicissimum*, purified, and chemically converted to the 2-deoxyparaherquamide analogue.³⁸ This new spiroindole was released in July 2010 as a combination treatment with abamectin and is currently marketed under the brand name Startect for use as an anthelmintic in sheep.⁴²

Elucidating the biosynthesis of paraherquamide A has been an area of intense research for several years, largely by Williams et al.

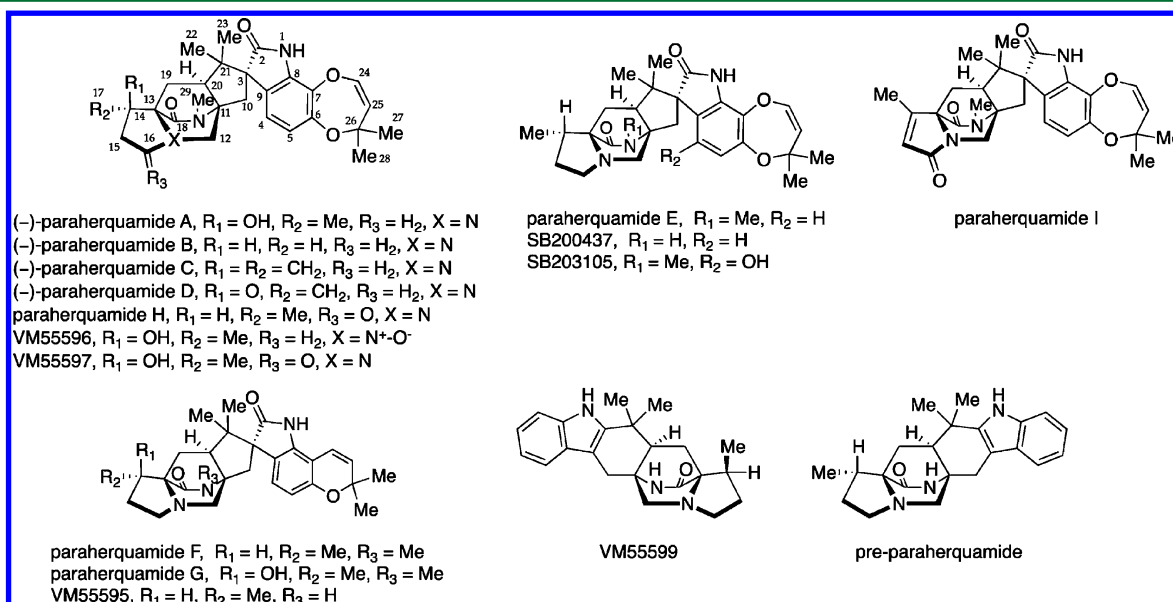


Figure 5. Structures of the paraherquamides and related compounds.

Table 6. Fungal Origin of the Paraherquamides

compound	fungal species	bioactivity	references
paraherquamide A	<i>P. charlesii</i> , ^a <i>P. cluniae</i> , <i>P. brasilianum</i> (= <i>P. paraherquei</i>), <i>Penicillium</i> sp. IMI 332995, ^a <i>Penicillium</i> species	anthelmintic antinematodal	4b–d, a, e, g, 38–42
paraherquamide B	<i>P. charlesii</i> , ^a <i>P. cluniae</i>	antinematodal	4b–d
paraherquamide C	<i>P. charlesii</i> ^a	antinematodal	4b, c, h
paraherquamide D	<i>P. charlesii</i> ^a	antinematodal	4b, c, h
paraherquamide E	<i>Aspergillus</i> sp. IMI 337664, <i>A. aculeatus</i> , <i>P. charlesii</i> , ^a <i>P. cluniae</i> , <i>Penicillium</i> sp. IMI 332995, ^a <i>Penicillium</i> species	antinematodal	4f, b, c, h, k, d, e, g
paraherquamide F	<i>P. charlesii</i> , ^a <i>Penicillium</i> sp. IMI 332995, <i>Penicillium</i> species	antinematodal	4b, c, h, e, g
paraherquamide G	<i>P. charlesii</i> , ^a <i>Penicillium</i> sp. IMI 332995, ^a <i>Penicillium</i> species	antinematodal	4b, c, h, e, g
paraherquamide H	<i>P. cluniae</i>		4d
paraherquamide I	<i>P. cluniae</i>		4d
VM55595	<i>Penicillium</i> sp. IMI 332995 ^a		4e
VM55596	<i>P. cluniae</i>	insecticidal	4d
VM55597	<i>P. cluniae</i> , <i>Penicillium</i> sp. IMI 332995 ^a	insecticidal	4d, e
VM55599	<i>P. paneum</i> , <i>Penicillium</i> sp. IMI 332995, ^a <i>A. japonicus</i> , ^b <i>P. charlesii</i> ^a		4i, k, e, j
SB200437	<i>Aspergillus</i> sp. IMI 337664 ^c	anthelmintic	4f
SB203105	<i>Aspergillus</i> sp. IMI 337664 ^c	anthelmintic	4f
pre-paraherquamide	<i>A. japonicus</i> , ^b <i>P. charlesii</i> ^a		4j

^aReclassified as a new species related to *Penicillium canescens*. ^bReclassified as *Aspergillus aculeatinus*. ^cReclassified as *Aspergillus fijiensis*.

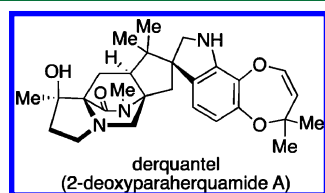
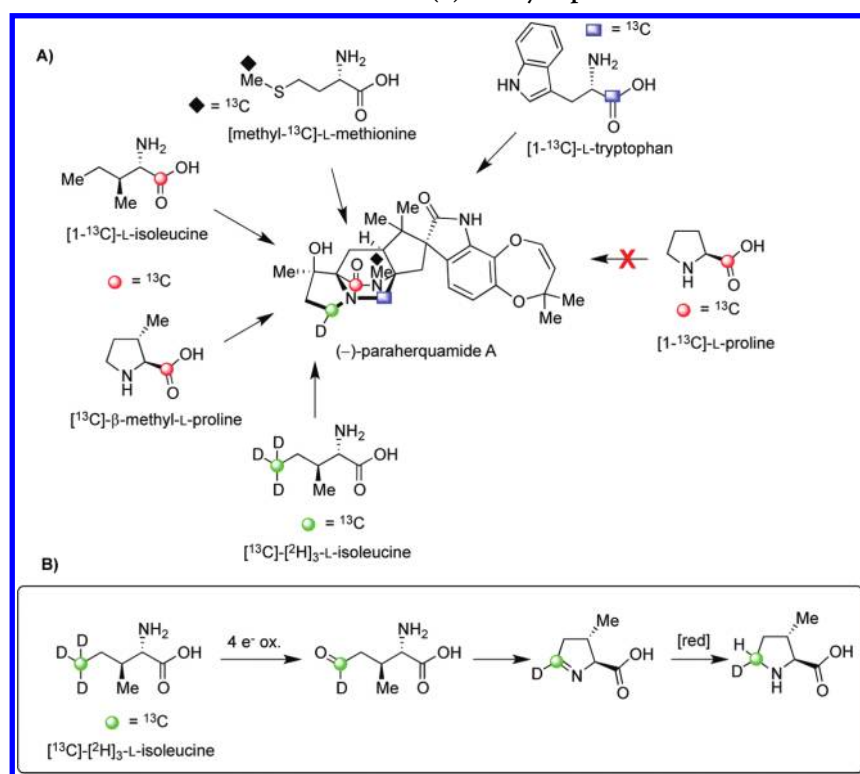


Figure 6. Structure of 2-deoxyparaherquamide A (derquantel).

On the basis of the results obtained from the precursor incorporation studies of the brevianamides, this group postulated

that the structurally similar paraherquamides follow a comparable biosynthetic pathway. To determine which amino acid residues are incorporated into paraherquamide A, feeding experiments were performed with *P. cf. canescens* using [1-¹³C]-L-tryptophan, [methyl-¹³C]-L-methionine, [1-¹³C]-L-proline, and [1-¹³C]-L-isoleucine (Scheme 9).⁴³ Through ¹³CNMR spectroscopy and electrospray mass spectrometry, the position and percentage of ¹³C incorporation were analyzed, and as expected, [1-¹³C]-L-tryptophan incorporated at the C-12 (paraherquamide numbering) position of the oxindole ring with 2.5% incorporation. Moreover, similar to results obtained

Scheme 9. (A) ¹³C-Labeled Amino Acid Incorporation into Paraherquamide A in *Penicillium cf. canescens*; (B) Proposed Biosynthetic Pathway for the Conversion of L-Isoleucine to 3(S)-Methyl-L-proline⁴³



in the marcfortine biosynthesis, a small amount (0.6%) of [*methyl*- ^{13}C]-L-methionine was found to incorporate at the nitrogen methyl of the monoketopiperazine ring.

With regard to the biogenesis of the β -methyl- β -hydroxyproline ring, precedent established by Arigoni and co-workers in the biosynthesis of bottromycin revealed incorporation of [*methyl*- ^{13}C]-L-methionine into the natural product at the 3(*R*)-methyl group of the β -methylproline ring of bottromycin.^{43b} However, results from the precursor incorporation study with *P. cf. canescens* showed that neither [*methyl*- ^{13}C]-L-methionine nor [$1\text{-}^{13}\text{C}$]-L-proline incorporated into the β -methyl- β -hydroxyproline ring of paraherquamide A. It was later found that [$1\text{-}^{13}\text{C}$]-L-isoleucine exhibited incorporation to the extent of 3.3–3.7%, with the label at the C-18 position. To establish a potential mechanism of incorporation of isoleucine into paraherquamide A, [^{13}C]-[^2H]₃-L-isoleucine was synthesized and fed to cultures of *P. cf. canescens*.^{43c,d} Results showed that only one deuterium label was retained, thus confirming a four-electron oxidative cyclization pathway to form β -methyl-L-proline (Scheme 9B).

This possibility was supported by an additional ^{13}C -labeling study, which revealed a higher percentage of ^{13}C incorporation in the monoketopiperazine ring when L-isoleucine underwent oxidative cyclization to form the β -methyl-L-proline prior to the precursor incorporation study with *P. cf. canescens*. An incorporation of 14.6% was observed in this case, compared to the previously observed 3.7% incorporation of L-isoleucine.^{43c,d}

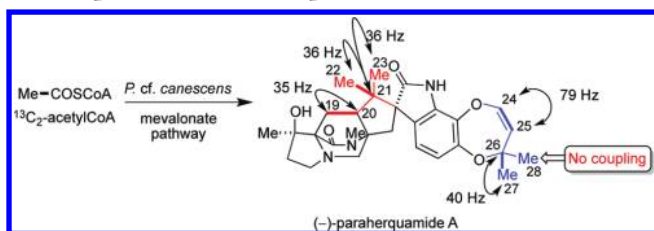
Additional biosynthetic research, regarding the exact origin of the two isoprene units present in paraherquamide A, has

been carried out and rigorously analyzed by Williams and colleagues. By providing [$^{13}\text{C}_2$]-acetate to *P. cf. canescens* it was demonstrated that the isoprene units in paraherquamide A were introduced in two distinct ways (Scheme 10).^{11,12} Through ^{13}C NMR spectroscopy, the coupling of intact C_2 units revealed that in the C_5 fragment formed by C-19 to C-23, C-19 was coupled to C-20, while C-21 was coupled to C-22 and C-23, but not to both simultaneously. For the second C_5 unit in paraherquamide A, C-24 to C-28, the coupling constants showed that C-24 was coupled to C-25, while C-26 was coupled to C-27, but not to C-28.^{11,12} On the basis of the observed coupling patterns, it was concluded that the introduction of the C-24 to C-28 isoprene unit must proceed through a direct prenyl transfer in which nucleophilic displacement occurs at the pyrophosphate-bearing methylene carbon, while the hydrophobic tail of DMAPP is buried in the enzyme active site.

The mechanistic formation of the C-19 to C-23 fragment was further complicated by the observed scrambling of the ^{13}C labels of the geminal methyl groups at C-22 and C-23, thus indicating a loss of stereochemical integrity during the biosynthetic construction of the quaternary center.^{11,12} These unexpected results pointed to a non-face-selective reverse prenyl transferase. In this case, the olefinic π system of DMAPP is presented in such a manner that both faces are susceptible to attack by the 2-position of the indole moiety. As shown in Scheme 11, one possible explanation is that the pyrophosphate group is anchored in the enzyme active site with the hydrophobic isopropenyl moiety being presented in a conformationally flexible disposition ($\text{A} \rightleftharpoons \text{B}$).

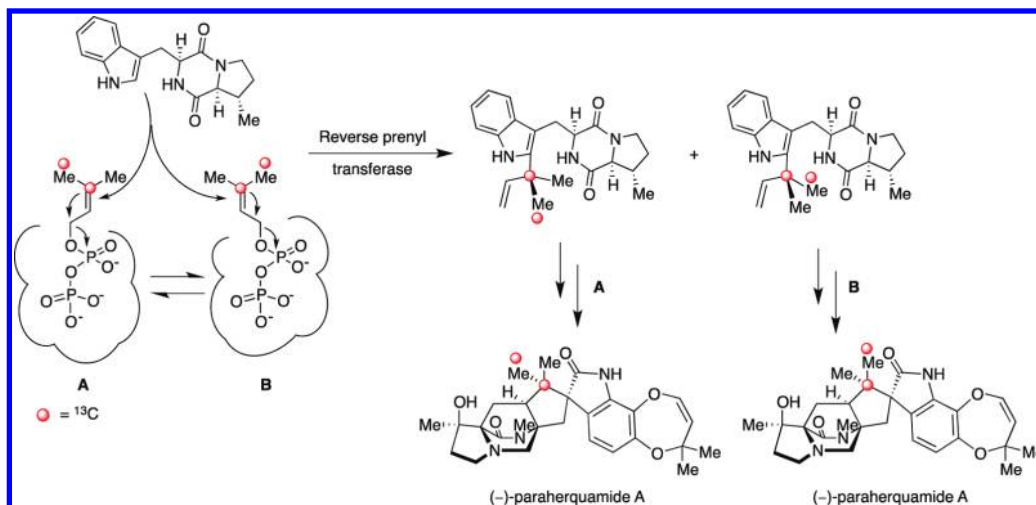
As shown in Scheme 12, the advanced hexacyclic indole derivative, ^{13}C -labeled pre-paraherquamide, was synthesized and provided to *P. cf. canescens* in a precursor incorporation study.⁴⁴ Results showed that this material incorporated with the double label intact and ^{13}C -incorporation of 0.72%, suggesting that pre-paraherquamide is in fact a pathway metabolite in paraherquamide biosynthesis. This information also indicates that formation of the dioxepin ring and the spiro-oxindole moiety occurs after the formation of this intermediate. Furthermore, the dioxopiperazine version of pre-paraherquamide (8) was also synthesized as an isotopomer and fed to *P. cf. canescens*; however, no incorporation was detected. Therefore, reduction of the tryptophan-derived carbonyl residue to the

Scheme 10. Relevant Coupling Patterns Observed in the ^{13}C NMR Spectrum of Paraherquamide A^a



^aThick lines represent intact acetate units.¹²

Scheme 11. Proposed Biosynthetic Mechanism of Reverse Prenylation in Paraherquamide A to Explain the Scrambling of the Geminal Methyl ^{13}C Labels Observed during the Precursor Incorporation Study¹²



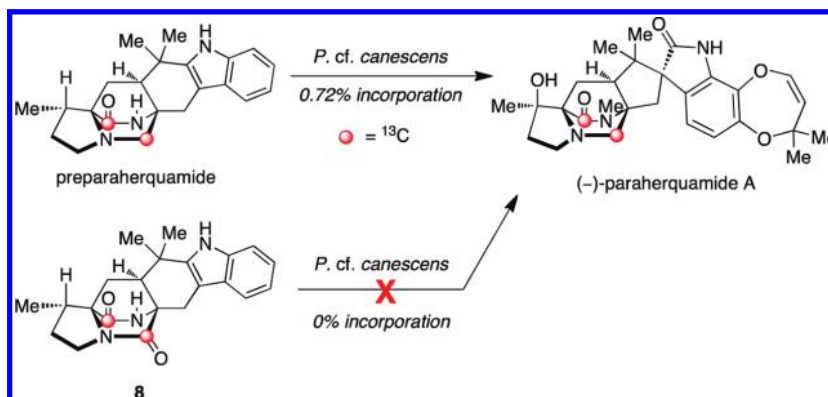
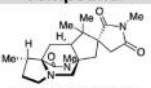
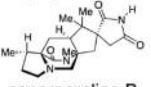
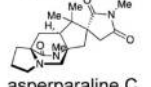
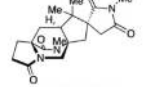
Scheme 12. Incorporation of ^{13}C -Labeled Pre-parahequamide into Parahequamide A in *P. cf. canescens*

Table 7. Fungal Origin and Bioactivity of the Asperparalines

compound	fungal species	bioactivity	reference
 asperparaline A	<i>A. japonicus</i> , ^a <i>Aspergillus</i> sp. IMI 337664 ^b	paralytic to silkworms, Anthelmintic	7a-b, 7c
 asperparaline B	<i>A. japonicus</i> ^a	paralytic to silkworms	7a-b
 asperparaline C	<i>A. japonicus</i> ^a	paralytic to silkworms	7a-b
 SB202327	<i>Aspergillus</i> sp. IMI 337664 ^b	anthelmintic	7c

^aReclassified as *Aspergillus aculeatinus*. ^bThis strain is probably *Aspergillus fijiensis*, which has pink sclerotia like IMI 337664.^{46f}

monoketopiperazine ring system must occur prior to formation of the core bridging bicycle.⁴⁴ These results are also consistent with the similar observation discussed above, regarding the biosynthesis of malbrancheamide.

Taken together, the data indicate that the subfamilies of alkaloids bearing a monoketopiperazine ring system might derive from a nonribosomal peptide synthase (NRPS) that reductively cleaves the dipeptide from the thiolation domain, resulting in off-loading of the tryptophan-derived carboxyl group as an aldehyde.¹ Ring closure of the cyclic amino acid secondary amine on this aldehyde results in imine formation followed by tautomerization to an azadiene (see Scheme 3). The azadiene intermediate then undergoes an IMDA to furnish the bicyclo[2.2.2]diazaoctane ring system. Conversely, the subfamilies that are comprised of a dioxopiperazine-based bicyclo[2.2.2]diazaoctane ring core would arise via an oxidative process to generate an azadiene species (see Scheme 6).

■ ASPERPARALINES

Asperparaline A is an indole-derived alkaloid containing the core bridging bicycle that was first isolated in 1997 by Hayashi and co-workers from the fungus *A. japonicus* JV-23, in Sakai, Japan.^{7a} However, the group of uniseriate black *Aspergillus* species has been extended from one species to seven species, and their chemical profiling has shown that the actual asperparaline producers are *A. aculeatinus* and *A. fijiensis* (Table 7).^{45,46} In 2000, the same group reported the isolation of two new related indole alkaloids, asperparalines B and C, from the same

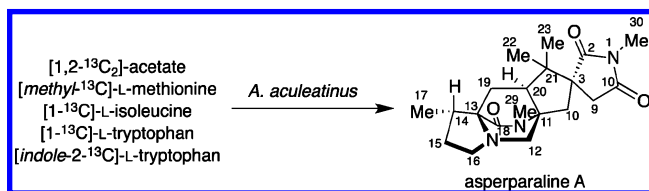
fungal strain.^{7b} In separate work, Everett et al. reported the isolation of asperparaline A, named aspergillimide, from *Aspergillus* sp. IMI 337664.^{7c} From the same fungal culture, most likely the new species *A. fijiensis*, the 16-keto derivative of asperparaline A was isolated and named SB202327. From a structural point of view, the asperparalines possess the core bicyclo[2.2.2]diazaoctane ring system. In addition, asperparalines A and B both contain a β -methylproline ring, whereas asperparaline C is the C-3-desmethyl analogue. The asperparalines also contain a unique 3-spiro-succinimide ring system, the orientation of which is consistent with the relative configuration of the spiro-oxindole ring system present in the parahequamides.⁴⁷

All members of the asperparaline family have exhibited some form of biological activity. Asperparaline A exhibited paralysis against silkworms at a dose of 10 $\mu\text{g/g}$ of the diet within 1 h and lasted for 7 to 10 h upon oral administration.⁷ Asperparalines B and C displayed almost the same effect on silkworms. Separately, asperparaline A and SB202327 were tested for anthelmintic activity against adult *Trichostrongylus colubriformis* infection in gerbils.⁷ Results from this experiment showed that asperparaline A displayed a 44% and 98% reduction in fecal egg count when dosed at 10 and 20 mg/kg, respectively. SB202327 exhibited in vitro activity against *Hemonchus contortus* L₃ larvae, but did not display in vivo activity against *T. colubriformis* in gerbils.

Since the asperparaline carbon skeleton closely resembles the parahequamides, Williams and co-workers directed their attention toward elucidating the biosynthesis of the asperparalines.

To determine the primary amino acid building blocks of asperparaline A [1,2- $^{13}\text{C}_2$]-acetate, [methyl- ^{13}C]-L-methionine, [1- ^{13}C]-L-isoleucine, [1- ^{13}C]-L-tryptophan, and [indole-2- ^{13}C]-L-tryptophan were employed in precursor incorporation studies (Scheme 13).⁴⁷ The position of the ^{13}C enrichment in

Scheme 13. Asperparaline A Precursor Incorporation Study



asperparaline A was determined by $^{13}\text{CNMR}$, and the percentage of labeled amino acid enrichment was determined by $^{13}\text{CNMR}$ and electrospray mass spectrometry. Results showed that [1,2- $^{13}\text{C}_2$]-acetate incorporates into asperparaline A in an analogous manner to that of paraherquamide A, indicating that the isoprene units in asperparaline A also arise from the mevalonic acid pathway. A significant amount of incorporation from [methyl- ^{13}C]-L-methionine was observed at the C-29 (25.2%) and C-30 (26.9%) positions of asperparaline A, but no incorporation was observed in any part of the β -methylproline ring. However, as expected, [1- ^{13}C]-L-isoleucine exhibited a 5.8% incorporation at the C-18 position of the β -methylproline ring, which indicates that the β -methylproline ring in both asperparaline A and paraherquamide A is biosynthesized along a similar, if not identical route. While the asperparalines appear at first glance to be the only prenylated indole alkaloids not derived from tryptophan, the precursor incorporation studies show otherwise. Both [1- ^{13}C]-L-tryptophan and [indole-2- ^{13}C]-L-tryptophan showed incorporation at the C-12 (6.4–7.2%) and C-2 (12.2%) positions of asperparaline A,

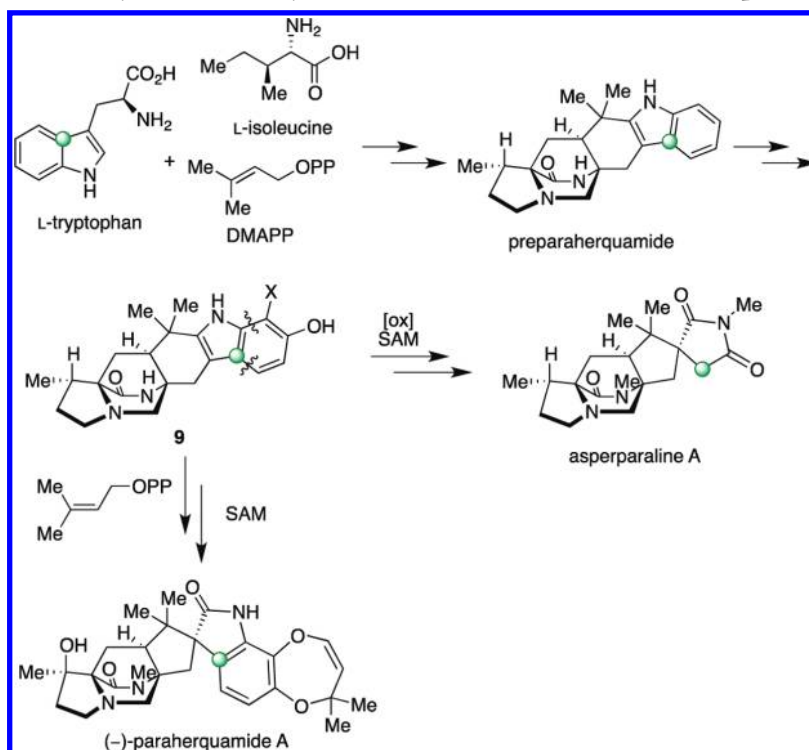
respectively. This indicated that tryptophan is responsible for the construction of the spiro-succinimide ring found in the asperparalines.

Results from the precursor incorporation studies, along with the recent detection of pre-paraherquamide in *A. aculeatinus*, provide indirect support that (a) pre-paraherquamide is a precursor to asperparaline A and (b) paraherquamide A and asperparaline A share a unified biosynthetic scheme.⁴¹ To address both of these hypotheses, a unified biosynthetic pathway has been proposed. As illustrated in Scheme 14, the proposed unified biogenesis of paraherquamide A and asperparaline A arises from pre-paraherquamide, which is derived from the common precursors L-tryptophan, L-isoleucine, and DMAPP.^{44,47} From pre-paraherquamide, the biosynthesis of paraherquamide A could arise via oxidation of the aromatic ring of pre-paraherquamide to the catechol derivative (9), followed by prenylation, dioxepin formation, and oxidation to the spiro-oxindole.¹² Along this same pathway, formation of the spiro-succinimide ring of asperparaline A would occur via interruption of the paraherquamide A biosynthesis just before the prenylation step of the catechol derivative.⁴⁷ This would provide a possible branching point for the oxidative cleavage of four carbons of the aromatic ring, thus resulting in the formation of the spiro-succinimide moiety.

■ STEPHACIDINS AND NOTOAMIDES

The stephacidins and notoamides comprise a family of prenylated indole alkaloids closely related to the paraherquamides. The first member to be isolated of what was to become the stephacidin and notoamide families was sclerotiamide (Figure 7). This was identified in 1996 from *Aspergillus sclerotiorum* by Gloer and White and was actually classified as a paraherquamide-type compound, despite containing a diketopiperazine as opposed to a monoketopiperazine, as all paraherquamide family members are constituted.^{5a} Structurally similar to the paraherquamides, sclerotiamide contains the core bicyclo[2.2.2]diazaoctane

Scheme 14. Proposed Unified Biosynthetic Pathway with Branch Point to Generate Paraherquamides or Asperparalines⁴⁷



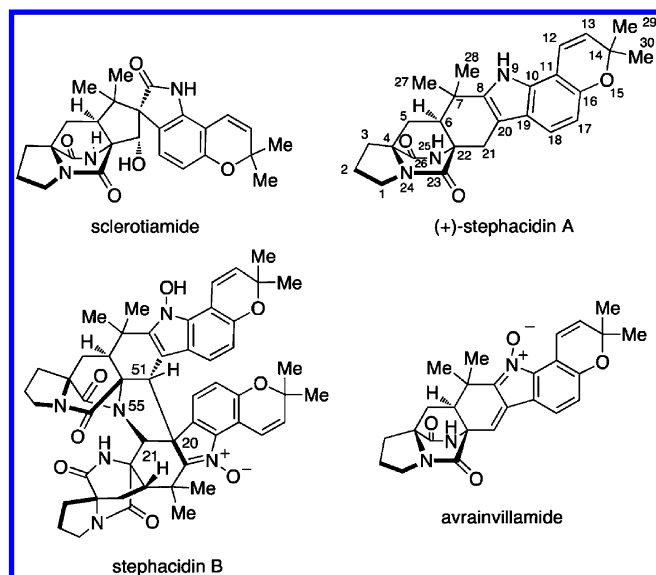


Figure 7. Structures of the stephacidins and related compounds.

moiety, a spiro-oxindole ring system with a pyran group, and an unsubstituted proline ring. However, unlike the paraherquamides, sclerotiamide retains an unreduced dioxopiperazine unit

and also contains a unique C-10 hydroxyl group (paraherquamide numbering). At the time, the unusual structural skeleton of sclerotiamide made it appear to be a member of the paraherquamides.

It was not until 1999 that other structurally related prenylated indole alkaloids were isolated, including (+)-stephacidin A, stephacidin B, and avrainvillamide.^{5b,c} Stephacidins A and B were both isolated from the terrestrial-derived *Aspergillus ochraceus* WC76466 by a group at Bristol-Myers Squibb,^{5b,c} while avrainvillamide was isolated independently from a marine-derived *Aspergillus* sp. CNC358 by Fenical and associates.^{5d} Two years later, Sugie and co-workers reported the isolation of avrainvillamide, under the name of CJ-17,665, from *A. ochraceus* CL41582.^{5e}

In 2007, Tsukamoto and co-workers isolated notoamides A–D (Figure 8) from a marine-derived *Aspergillus* sp. MF297-2 found growing on the mussel *Mytilus edulis*, collected off the Noto Peninsula in the Sea of Japan.^{6a} Along with the notoamides, the known compounds sclerotiamide, (+)-stephacidin A, and deoxybrevianamide E were also obtained. Notoamides A–D contain the pyranoidole ring system common in the stephacidins and some of the paraherquamides. Notoamides A and B both contain the core bridging bicycle with C-12 oxidation (paraherquamide numbering), as well as the spiro-oxindole ring system. In

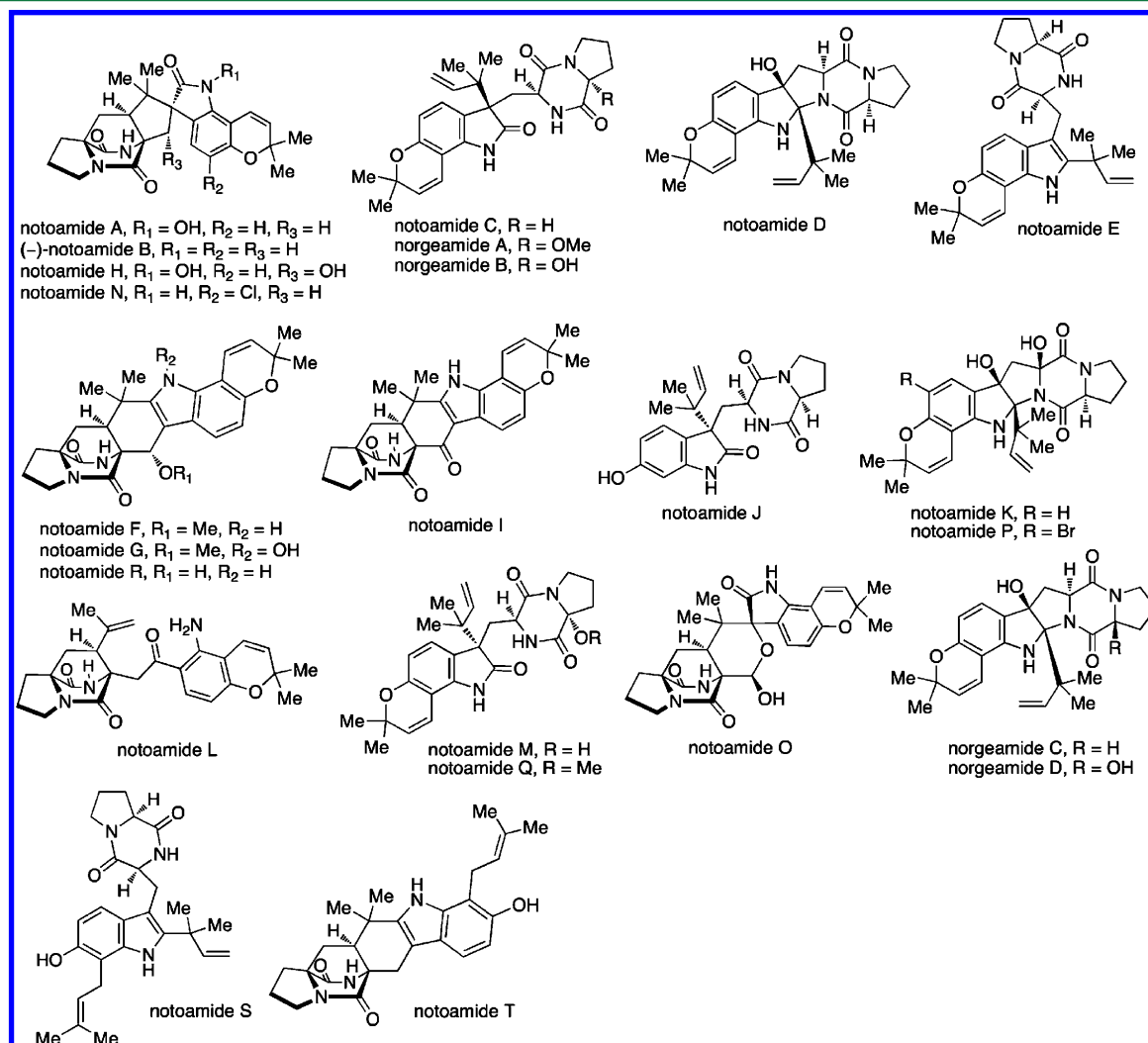


Figure 8. Structures of the notoamides and related compounds.

2005, a group of structurally similar indole alkaloids, named the norgeamides, was isolated at the Hans-Knöll Institute from a cold water *Aspergillus* sp.⁴⁸ Minor metabolites of the notoamides continue to be isolated from marine-derived *Aspergillus* sp. MF297-2; notoamides E–R are known natural products, while notoamides S and T are potential biosynthetic intermediates that have yet to be isolated.^{2,6b–e,49}

Stephacidin A and notoamide B were also isolated recently from the terrestrial fungus *Aspergillus versicolor*, along with a new metabolite, versicolamide B.^{6f} This new secondary metabolite displays the rare *anti*-relative configuration at C-6 (stephacidin numbering) in the bicyclo[2.2.2]diazaoctane ring system, which is similar to that in brevianamides A and B. Following the isolation of versicolamide B from *A. versicolor*, it was found that this metabolite is also produced by the marine-derived *Aspergillus* sp. MF297-2.^{6d} Further investigation of the metabolites isolated from *A. versicolor* and *Aspergillus* sp. MF297-2 led to the discovery of another unique trait within this family. (–)-Stephacidin A, (+)-notoamide B, and (+)-versicolamide B from *A. versicolor* are the corresponding enantiomers to (+)-stephacidin A, (–)-notoamide B, and (–)-versicolamide B produced by the marine-derived *Aspergillus* sp. MF297-2 (Figure 9).^{6f} Additionally, (–)-stephacidin A is

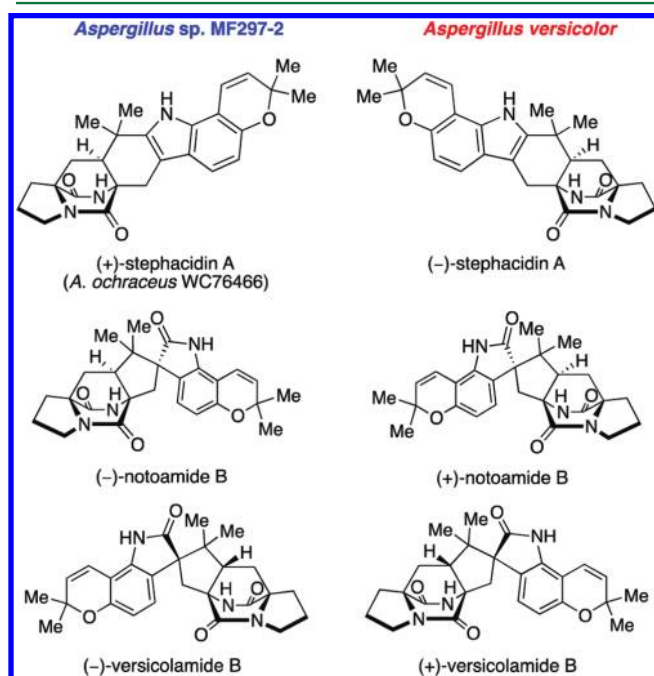


Figure 9. Structures of the enantiomeric metabolites produced by *Aspergillus* sp. MF297-2 and *A. versicolor*.

the enantiomer of (+)-stephacidin A isolated from *A. ochraceus* WC76466, a terrestrial fungus. The discovery of these enantiomeric metabolites produced by two different *Aspergillus* species represents the first known occurrence of reverse prenylated enantiomeric natural products.

As shown in Table 8, numerous prenylated indole alkaloids belonging to the strophacidin and notoamide family have been isolated from various *Aspergillus* fungal cultures. Taxonomically, the notoamides (stephacidins, aspergamides, avrainvillamides, sclerotiamide, versicolamide, norgeamides) are produced by two groups of *Aspergillus*: species in sections *Circumdati* and *Versicolores*. In section *Circumdati*, *A. sclerotiorum*,^{5a} *A. ochraceus*,^{5b–d} *A. ostianus*,^{6g} and *A. westerdijkiae*⁵⁰ can produce

Table 8. Fungal Origin of the Stephacidins and Notoamides

compound	fungal species	bioactivity	reference
sclerotiamide	<i>A. sclerotiorum</i> , <i>Aspergillus</i> sp. MF297-2 ^a	insecticidal antifeedant	5a, 6a
(+)-stephacidin A	<i>A. ochraceus</i> WC76466, <i>Aspergillus</i> sp. MF297-2 ^a	cytotoxic	5b, c, 6a
(–)-stephacidin A	<i>A. versicolor</i>		6f
stephacidin B	<i>A. ochraceus</i> WC76466	cytotoxic	5b, c
avrainvillamide	<i>A. ochraceus</i> CL41582, <i>Aspergillus</i> sp. CNC358	antibiotic cytotoxic	5e, d
notoamide A	<i>Aspergillus</i> sp. MF297-2 ^a		6a
(–)-notoamide B	<i>Aspergillus</i> sp. MF297-2 ^a		6a
(+)-notoamide B	<i>A. versicolor</i>		6f
notoamide C	<i>Aspergillus</i> sp. MF297 ^a		6a
notoamide D	<i>Aspergillus</i> sp. MF297 ^a		6a
notoamide E	<i>Aspergillus</i> sp. MF297 ^a		6b
notoamide F	<i>Aspergillus</i> sp. MF297, ^a <i>A. ostianus</i>		6c, g
notoamide G	<i>Aspergillus</i> sp. MF297 ^a		6c
notoamide H	<i>Aspergillus</i> sp. MF297 ^a		6c
notoamide I	<i>Aspergillus</i> sp. MF297 ^a		6c
notoamide J	<i>Aspergillus</i> sp. MF297 ^a		6c
notoamide K	<i>Aspergillus</i> sp. MF297 ^a		6c
notoamide L	<i>Aspergillus</i> sp. MF297 ^a		6d
notoamide M	<i>Aspergillus</i> sp. MF297 ^a		6d
notoamide N	<i>Aspergillus</i> sp. MF297 ^a		6d
notoamide O	<i>Aspergillus</i> sp. MF297 ^a		6e
notoamide P	<i>Aspergillus</i> sp. MF297 ^a		6e
notoamide Q	<i>Aspergillus</i> sp. MF297 ^a		6e
notoamide R	<i>Aspergillus</i> sp. MF297, ^a <i>A. ostianus</i>		6e, g
(–)-versicolamide B	<i>Aspergillus</i> sp. MF297 ^a		6d
(+)-versicolamide B	<i>A. versicolor</i>		6f
norgeamide A	<i>Aspergillus</i> species	cytotoxic	48, 51
norgeamide B	<i>Aspergillus</i> species	cytotoxic	48, 51
norgeamide C	<i>Aspergillus</i> species	cytotoxic	48, 51
norgeamide D	<i>Aspergillus</i> species, <i>A. versicolor</i>	cytotoxic	48, 6f, 51

^aUndergoing reidentification.

notoamides. In addition, chemical screening has shown that *A. melleus*, *A. persii*, *A. petrakii*, and *A. bridgeri* can also produce notoamides, while the remaining 12 species in section *Circumdati* do not appear to produce them. In *Aspergillus* section *Versicolores*, a strain of *A. versicolor* can produce the stereoisomers, as compared to those produced by species in section *Circumdati* of the notoamides.^{6f} We have detected these compounds later in other strains of *A. versicolor* and in two strains of the closely related species *Aspergillus protuberus* (Frisvad, Unpublished).

Biologically, the notoamides appear to have moderate and limited biological activity, while the structurally related norgeamides have all been shown to inhibit the growth of a variety of cancer cell lines with varying efficacies.⁵¹ Preliminary biological assays revealed that norgeamide A is the most cytotoxic against several carcinoma cell lines. However, sclerotiamide, avrainvillamide, and strophacidins A and B appear to be the most bioactive of the strophacidins and notoamides. These secondary metabolites display anti-insecticidal, antibacterial, and cytotoxic properties.^{5a,c,e}

Sclerotiamide has been shown to exhibit potent activity against first instar larvae of the corn earworm *Helicoverpa zea*.^{5a} When incorporated into standard test diets of 200 ppm, sclerotiamide resulted in a 46% mortality rate of *H. zea*. There was also a 98% reduction in growth rate, relative to controls,

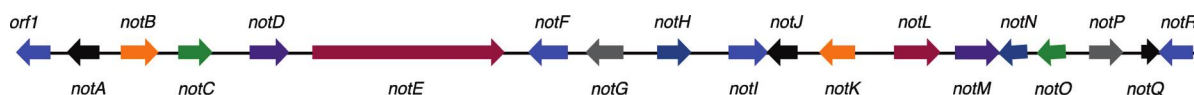
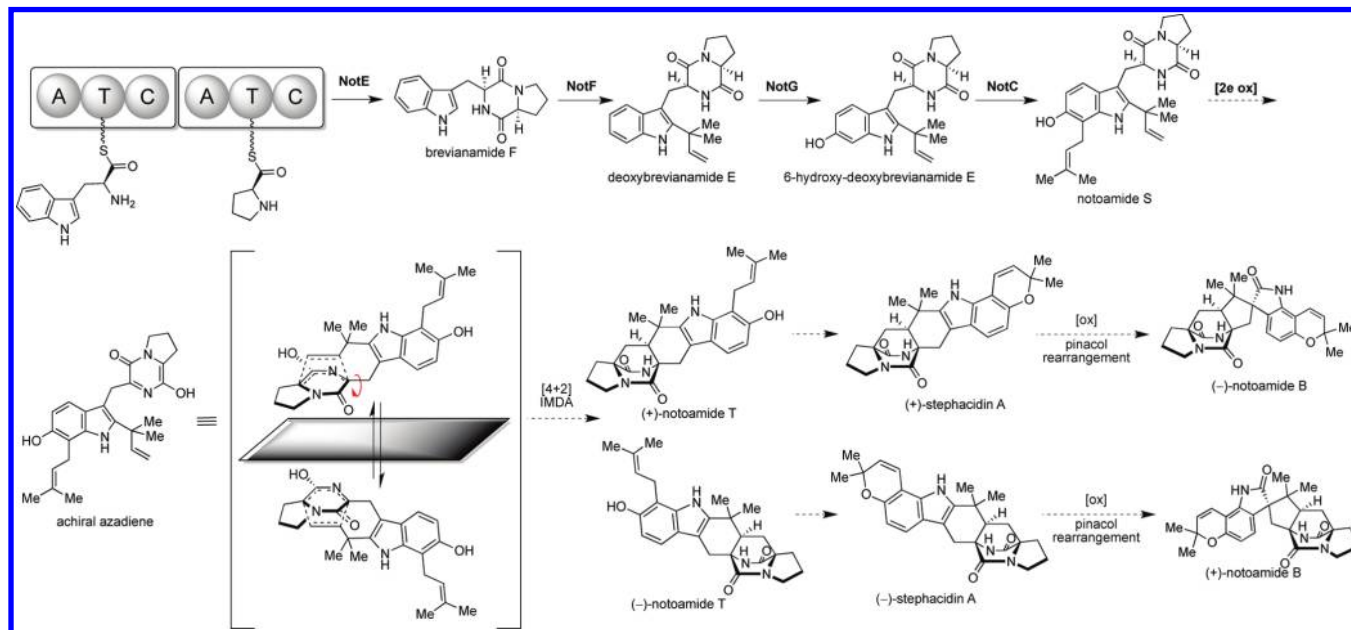


Figure 10. Notoamide (*not*) biosynthetic gene cluster from *Aspergillus* sp. MF297-2.

Scheme 15. Currently Postulated Biosynthetic Pathway of the Stephacidins and Notoamides in *Aspergillus* sp. MF297-2 and *A. versicolor*



among the survivors. Several physiological effects, such as shriveling and blackening, were also observed in the larvae. In a feeding assay against adults and larvae of the fungivorous beetle *Carpophilus hemipterus*, at a 100 ppm dietary level, sclerotiamide reduced feeding rates by 44% and 40%, respectively.

Avrainvillamide is an antibiotic that inhibits the growth of multi-drug-resistant *Staphylococcus aureus*, *Streptococcus pyogenes*, and *Enterococcus faecalis* with MICs of 12.5, 12.5, and 25 $\mu\text{g/mL}$, respectively.^{5e} When tested against *Escherichia coli*, at 100 $\mu\text{g/mL}$ it showed no antibacterial activity. Avrainvillamide also displays cytotoxic activity against a variety of tumor cell lines, including human colon HCT116 cell lines, melanoma MALME-3 M cell lines, and two breast cancer cell lines, $\beta\text{T-549}$ and T-47D.^{5d}

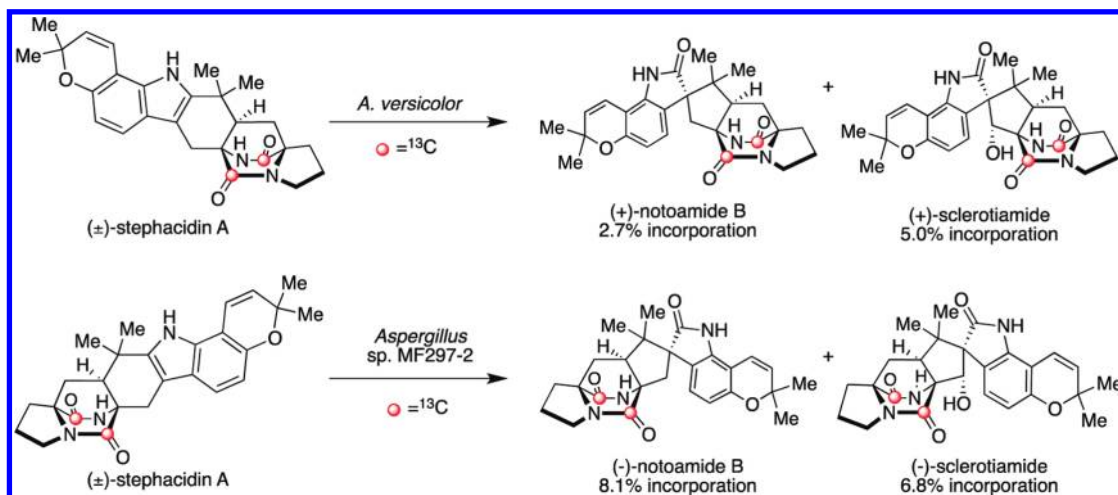
Stephacidins A and B both demonstrate cytotoxicity against various human tumor cell lines; however, stephacidin B exhibited more potent and selective antiproliferative activities, especially for testosterone-dependent LNCaP cells.^{5b,c} The effects of stephacidins A and B indicate a novel mode of action since they are not mediated by p53, *mdr*, or *bc12*, and they are not tubulin- or topoisomerase II-mediated. Recent studies conducted by Myers et al. suggest that the growth-inhibitory activity of stephacidin B is a result of retrodimerization to form the monomer avrainvillamide, which may then act as a novel Michael acceptor via the α,β -unsaturated nitrone group.⁵²

Following the isolation of the first set of enantiomeric reverse prenylated indole metabolites, attention has been directed toward elucidating the biosynthesis of the stephacidins and notoamides. In collaboration with the Sherman laboratory at the University of Michigan, additional insight into the biogenesis of the stephacidins and notoamides was obtained through genome sequencing of *Aspergillus* sp. MF297-2 and *A. versicolor*.^{1,53} Through data mining and deep annotation, the

stephacidin and notoamide biosynthetic gene clusters have been identified and characterized. The target gene cluster in *Aspergillus* sp. MF297-2 was identified from an open reading frame (*orf*) named *notE*, using the fumitremorgin gene *ftmA* as an in silico probe.^{1,53a} Bioinformatics analysis revealed a total of 18 genes (*notA–notR*) encompassing ~44 kb of DNA (Figure 10), including a bimodular NRPS (NotE) and two aromatic prenyltransferases, *notC* and *notF*. These latter two genes were functionally expressed in *E. coli*, with NotC biochemically characterized as a normal prenyltransferase and NotF determined to be a reverse prenyltransferase.

On the basis of the above genome sequencing, bioinformatic mining, and biochemical studies, a comprehensive stephacidin and notoamide biogenesis was proposed (Scheme 15). Following the condensation of tryptophan and proline via NotE to form the common precursor brevianamide F, NotF is then responsible for the C-2 reverse prenylation of brevianamide F to give deoxybrevianamide E.^{1,53} Subsequent oxidation at the C-6 indole position gives rise to 6-hydroxydeoxybrevianamide E, which has been confirmed recently to be the NotC substrate. The reaction of 6-hydroxydeoxybrevianamide E in the presence of NotC and DMAPP affords 6-hydroxy-7-prenyldeoxybrevianamide E, which we have named notoamide S.

It is the oxidation and tautomerization of notoamide S that we have proposed presently to give rise to the putative achiral azadiene intermediate, which undergoes an enantiospecific IMDA reaction in the respective *Aspergillus* fungal cultures to provide the proposed metabolite, notoamide T. Oxidative cyclization of (+)-notoamide T in *Aspergillus* sp. MF297-2 would afford (+)-stephacidin A, which, following an oxidation and pinacol-type rearrangement to the spiro-oxindole, yields (–)-notoamide B. The enantiomerically opposite IMDA reaction in *A. versicolor*

Scheme 16. (\pm)-[^{13}C]₂-Stephacidin A Precursor Incorporation Study with *Aspergillus versicolor* and *Aspergillus* sp. MF297-2.⁵⁴

would yield (–)-notoamide T to then furnish (–)-stephacidin A and (+)-notoamide B.

Recent precursor incorporation studies carried out by Williams et al. have provided further validation for the proposed biogenesis of the stephacidins and notoamides. More specifically, (\pm)-[^{13}C]₂-stephacidin A was synthesized and provided to the marine-derived *Aspergillus* sp. MF297-2 and the terrestrial-derived *A. versicolor* (Scheme 16).⁵⁴ Analysis of the metabolites revealed enantiospecific incorporation of (+)-stephacidin A into (–)-notoamide B and (–)-sclerotiamide in *Aspergillus* sp. MF297-2, and (–)-stephacidin A into (+)-notoamide B and (+)-sclerotiamide in *A. versicolor*. These results indicate the presence of stereospecific yet complementary face-selective indole oxidases in the two fungal cultures.

■ RELATED METABOLITES LACKING THE BICYCLO[2.2.2]DIAZAOCTANE RING SYSTEM

From a structural point of view, the austamides,²⁷ citrinadins,⁵⁵ and the PF1270 family of metabolites⁵⁶ appear to be closely related to the previously discussed indole alkaloids, albeit lacking the core bicyclo[2.2.2]diazaoctane nucleus. As shown in Table 9, these related families of metabolites have been isolated from various *Penicillium* and *Aspergillus* species and also display a plethora of bioactivity. From a taxonomic perspective, the citrinidins are produced by *Penicillium citrinum*,⁵⁵ and the closely related PF1270 compounds are produced by *Penicillium waksmanii*;⁵⁶ however it is more likely that *Penicillium westlingii* produces both families of compounds, as this species is closely related to *P. citrinum*.⁵⁷ *P. westlingii* was treated as a synonym of *P. waksmanii* by Pitt,³⁷ but has later been shown to be a valid species.^{35a,57} *Talaromyces thermophilus* strain YM 1-3 produces talathermophilins A and B,⁵⁸ secondary metabolites related to notoamide E, except that glycine or alanine is replacing proline and the end products have a double bond conjugating one of the ketones in the dioxopiperazine to the indole part of the molecule. We have found that the glycine and alanine notoamide E analogues (without the extended conjugation) are also formed in *T. thermophilus* CBS 116.72, albeit accumulated in much smaller amounts compared to the end products. *T. thermophilus* has recently been shown to be closely related to *Thermomyces* and will probably be transferred to that genus.^{61c,d} This means that no strains of *Talaromyces* sensu stricto have been reported to

produce compounds with the core bicyclo[2.2.2]diazaoctane nucleus yet.

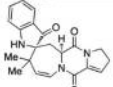
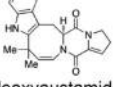
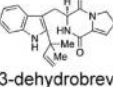
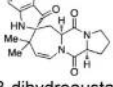
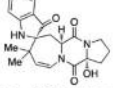
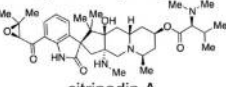
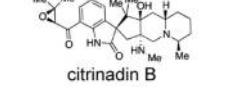
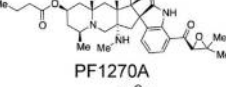
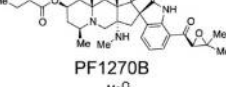
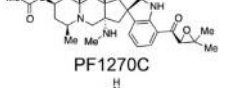
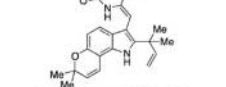
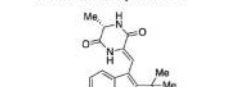
Structurally, austamide, 12,13-dihydroaustamide, and 12,13-dihydro-12-hydroxyaustamide all contain a spiro-indoxyl group similar to the brevianamides; however, these metabolites also contain a unique seven-membered spiro-indoxyl ring.²⁷ Another metabolite isolated from *A. pseudoustus* is deoxy-austamide, which displays an eight-membered ring between the indole and dioxopiperazine ring, making it a potential metabolic precursor to the austamides. Furthermore, deoxybrevianamide E, the known biosynthetic precursor to the brevianamides, stephacidins, and notoamides, was also isolated from austamide-producing *A. pseudoustus*, thus indicating the possibility of a shared biogenesis between the brevianamides and austamides.²⁷

Citrinadins A and B were isolated from the fungus *P. citrinum*, which was found growing on the red alga *Actinotrichia fragilis*.⁵⁵ These metabolites appear to be structurally related to the marcfortine and chrysogenamide families of indole alkaloids, in that the citrinadins also contain a substituted pipecolate moiety, C-7 indole substitution, and a spiro-oxindole ring system. While the citrinadins do not contain the core bridging bicyclo[2.2.2]diazaoctane, Kobayashi and co-workers proposed that they could biosynthetically arise from a marcfortine-type skeleton.^{55b} As shown in Figure 11, loss of the amide carbonyl carbon (C-18) followed by oxidation at C-15 and C-20 and methylation at C-17 could lead to an advanced citrinadin metabolite.

The PF1270 family of metabolites was isolated from a soil-borne fungal culture, *P. waksmanii* = *P. westlingii*, in Japan.⁵⁶ The carbon skeleton of these metabolites is closely related to the citrinadins, as well as the marcfortine and chrysogenamide families. Similar to the citrinadins, PF1270A, PF1270B, and PF1270C contain a substituted pipecolic acid residue, a C-7-substituted indole ring, and a spiro-oxindole moiety. Biosynthetically, this family could also arise from a marcfortine-type compound following loss of the amide carbonyl carbon.

Recently, two novel metabolites, talathermophilins A and B, were isolated from the thermophilic fungus *T. thermophilus*.^{58a} These new metabolites have been postulated to serve as potential precursors to the more advanced metabolites containing the bridging bicycle. Unlike notoamide B, metabolites tentatively identified by HPLC-MS as dihydrotalathermophilins A and B appear to be efficiently converted into the talathermophilins A

Table 9. Fungal Origin of Several Related Secondary Metabolites

compound	fungal species	bioactivity	reference
 austamide	<i>A. ustus</i> ^a	toxic to ducklings	27
 deoxyaustamide	<i>A. ustus</i> , ^a <i>P. italicum</i>	toxic to ducklings	27b,d
 deoxy-12,13-dehydrobrevianamide E	<i>A. ustus</i> , ^a <i>P. italicum</i> , <i>P. ulaiense</i>	toxic to ducklings	27b,d,e, 22, 23
 12,13-dihydroaustamide	<i>A. ustus</i> ^a	toxic to ducklings	27b
 12,13-dihydro-12-hydroxyaustamide	<i>A. ustus</i> ^a	toxic to ducklings	27c
 citrinadin A	<i>P. citrinum</i> N059	cytotoxic	55
 citrinadin B	<i>P. citrinum</i> N059	cytotoxic	55b
 PF1270A	<i>P. waksmanii</i> PF1270 ^b	histamine H3R inhibitor	56
 PF1270B	<i>P. waksmanii</i> PF1270 ^b	histamine H3R inhibitor	56
 PF1270C	<i>P. waksmanii</i> PF1270 ^b	histamine H3R inhibitor	56
 talathermophilin A	<i>Talaromyces thermophilus</i>	antiparasitodal	58
 talathermophilin B	<i>Talaromyces thermophilus</i>	antiparasitodal	58

^aReclassified as *Aspergillus pseudoustus*. ^bProbably *Penicillium westlingii*, closely related to *P. citrinum*.

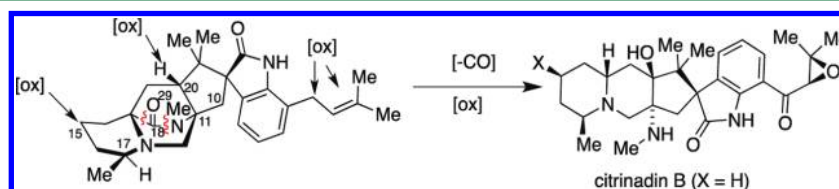


Figure 11. Possible biosynthetic precursor of citrinadins.

and B, which, according to metabolite profiling, appear to be the end products. Indeed the structures of dihydrotalathermophilins A and B were recently elucidated^{58b} and named talathermophilins

C and D. There were no indications of other metabolites with a bicyclo[2.2.2]diazaoctane ring system and/or any of the spiro-compounds of the notoamide type in *T. thermophilus*, where

Table 10. Production of Bicyclo[2.2.2]diazaoctane Compounds and Austamides in Different Ecological/Phylogenetic Series of *Aspergillus* and *Penicillium* (Frisvad, unpublished and refs 22–24, 26, 29, 45, 46, 50, based on the methods described in ref 63)

secondary metabolite group	section or series	species	production of bicyclo[2.2.2]diazaoctane compounds (and/or austamides)
brevianamides	<i>P. series Olsonii</i>	<i>P. brevicompactum</i>	+
		<i>P. astrolabium</i>	—
		<i>P. bialowiezense</i>	—
		<i>P. neocrassum</i>	—
		<i>P. olsonii</i>	—
brevianamides	<i>P. series Viridicata</i>	<i>P. viridicatum</i>	+
		<i>P. aurantiogriseum</i>	—
		<i>P. cyclopium</i>	—
		<i>P. freii</i>	—
		<i>P. melanoconidium</i>	—
		<i>P. neoechinulatum</i>	—
		<i>P. polonicum</i>	—
		<i>P. tricolor</i>	—
marcfortines	<i>P. series Roquefortorum</i>	<i>P. paneum</i>	+
		<i>P. carneum</i>	—
		<i>P. roqueforti</i>	—
		<i>P. psychrosexualis</i>	—
marcfortines	<i>Aspergillus</i> section <i>Flavipedes</i>	<i>A. carneus</i>	+
		<i>A. niveus</i>	—
		<i>A. flavipes</i>	—
chrysogenamides	<i>P. series Chrysogena</i>	<i>P. chrysogenum</i>	+
		<i>P. dipodomyis</i>	—
		<i>P. flavigenum</i>	—
		<i>P. nalgiovense</i>	—
		<i>P. rubens</i>	—
paraherquamides	<i>P. series Canescentia</i>	<i>P. cf. canescens</i>	+
		<i>P. canescens</i>	—
		<i>P. yarmokense</i>	—
		<i>P. janczewskii</i>	—
		<i>P. jensenii</i>	—
		<i>P. murcianum</i>	—
		<i>P. nigricans</i>	—
		<i>P. radiatolobatum</i>	—
		<i>P. brasiliense</i>	+
paraherquamides	<i>P. series Simplicissima</i>	<i>P. cluniae</i>	+
		<i>P. glaucoroseum</i>	+
		<i>P. janthinellum</i>	—
		<i>P. piscarium</i>	—
		<i>P. pulvillorum</i>	—
		<i>P. simplicissimum</i>	—
		<i>P. simplicissimum</i>	—
asperparalines	<i>Aspergillus</i> section <i>Nigri</i> series <i>Aculeati</i>	<i>A. aculeatinus</i>	+
		<i>A. fijiensis</i>	+
		<i>A. aculeatus</i>	—
		<i>A. indologenus</i>	—
		<i>A. japonicus</i>	—
		<i>A. uvarum</i>	—
		<i>A. violaceofuscus</i>	—
		<i>A. bridgeri</i>	+
notoamides, stephacidins, aspergamides, avrainvillamides	<i>Aspergillus</i> section <i>Circumdati</i>	<i>A. melleus</i>	+
		<i>A. ochraceus</i>	+
		<i>A. ostianus</i>	+
		<i>A. persii</i>	+
		<i>A. petrakii</i>	+
		<i>A. sclerotiorum</i>	+
		<i>A. westerdijkiae</i>	+
		<i>A. auricomus</i>	—
		<i>A. cretensis</i>	—
		<i>A. elegans</i>	—
		<i>A. flocculosus</i>	—
		<i>A. flocculosus</i>	—

Table 10. continued

secondary metabolite group	section or series	species	production of bicyclo[2.2.2]diazaoctane compounds (and/or austamides)
notoamides, opposite stereochemistry	<i>Aspergillus</i> series <i>Versicolores</i>	<i>A. insulicola</i>	—
		<i>A. muricatus</i>	—
		<i>A. neobridgeri</i>	—
		<i>A. ochraceopetaliformis</i>	—
		<i>A. pseudoelegans</i>	—
		<i>A. roseoglobulosus</i>	—
		<i>A. steynii</i>	—
		<i>A. sulphureus</i>	—
		<i>A. versicolor</i>	+
		<i>A. protuberus</i>	+
austamides and deoxybrevianamides	<i>Aspergillus</i> section <i>Usti</i>	<i>A. sydowii</i>	—
		<i>A. pseudoustus</i>	+
		<i>A. amylovorus</i>	—
		<i>A. calidoustus</i>	—
		<i>A. californicus</i>	—
		<i>A. carlsbadensis</i>	(brevianamide A)
		<i>A. cavernicola</i>	—
		<i>A. deflectus</i>	—
		<i>A. egyptiacus</i>	—
		<i>A. elongatus</i>	(possibly notoamide E)
		<i>A. germanicus</i>	—
		<i>A. granulosis</i>	—
		<i>A. heterothallicus</i>	—
		<i>A. insuetus</i>	—
		<i>A. kassunensis</i>	—
		<i>A. keveii</i>	—
		<i>A. lucknowensis</i>	—
		<i>A. monodii</i>	—
		<i>A. pseudodeflectus</i>	—
		<i>A. puniceus</i>	—
deoxybrevianamides	<i>P. series Italica</i>	<i>A. subsessilis</i>	—
		<i>A. turkensis</i>	—
		<i>A. ustus</i>	—
		<i>P. italicum</i>	+
		<i>P. ulaiense</i>	+

glycine or alanine analogues of the proline-derived metabolites discussed here could theoretically have been secondary metabolites in this themophilic species.

CHEMOTAXONOMIC OVERVIEW

The bioactive prenylated brevianamide F-derived secondary metabolites, some of which have the unique bicyclo[2.2.2]-diazaoctane ring system, have mostly been found in *Aspergillus* and *Penicillium*. The bicyclo[2.2.2]diazaoctane ring system has been found only in fungi that can produce prenylated tryptophan-proline (or pipecolic acid) dioxopiperazines and not in any other tryptophan-containing prenylated dioxopiperazines, which are also common in *Aspergillus* and *Penicillium*. However, *Malbranchea* is unrelated phylogenetically to *Penicillium* and *Aspergillus*, and thus the biosynthetic gene cluster for the bicyclo[2.2.2]diazaoctane ring system compounds could have been horizontally transferred or established through convergent evolution. Dioxopiperazines may have been some of the first secondary metabolites to have been produced by microorganisms during evolution, as they are of widespread occurrence and very bioactive.⁵⁹ More elaborate natural product structures, like those reviewed here, are of restricted taxonomic distribution. The genera *Penicillium* and *Aspergillus* are closely related, yet both contain species that can produce

optical enantiomers of the same or nearly the same molecule. For example, the genus *Aspergillus* can be subdivided into three major clades,⁶⁰ all of which contain producers of brevianamide F. In the *Neosartorya* clade, the end products based on brevianamide F (tryptophanyl-prolyl-dioxopiperazine) are the fumitremorgins and spirotryprostatins;²⁹ in the *Emericella* clade, *A. versicolor* produces (+)-versicolamide B; and, in the *Eurotium* clade, *A. sclerotiorum* produces (–)-versicolamide B. *Penicillium* can be roughly subdivided into six clades,^{61a} or 25 clades^{61b,c} of which only two contain producers of the bicyclo[2.2.2]-diazaoctane ring system secondary metabolites. Clade 5 contains *P. brasilianum*, *P. glaucoroseum*, and *P. cluniae*, all producing paraherquamides, while clade 6 contains all the remaining producers in *Penicillium*: *P. paneum* of marcfortines, *P. chrysogenum* of chrysogenamides, *P. brevicompactum* and *P. viridicatum* of brevianamides, and the less related *P. cf. canescens* of paraherquamides. The food-borne species in *Penicillium* subgenus *Penicillium* appear to represent a hot-spot for producers of the bicyclo[2.2.2]diazaoctane ring system compounds, but even within this subgenus, the compounds are widely distributed among the subclades, even as both *anti* (*P. chrysogenum*) and *syn* (the other species) compounds. The production of paraherquamides by soil-borne types of *Penicillium* in both clade 5 (*P. brasilianum*) and 6 (*P. cf. canescens*) could

indicate horizontal transfer of the whole biosynthetic gene cluster or reinventive evolution guided by selection. Thus, dioxopiperazines may have been vertically inherited from a common ancestor and may be also simple dioxopiperazine prenylation, but the end products of the biosynthetic routes appear to be unique and specific for the different species. Within closely related species the same secondary metabolites seem to be produced and could thus be derived from a common ancestor. For example, in *Aspergillus* section *Circumdati*, of the 20 different species known,⁵⁰ eight produce notoamides and 12 do not, but those that are produced are of the same chemical type. In section *Circumdati*, the notoamides are produced in two different clades, which are not sister clades.⁵⁰ The “central” clade contains *A. ochraceus*, *A. petrakii*, *A. ostianus*, *A. melleus*, and *A. westerdijkiae*,⁵⁰ all producing notoamides, while in a second clade consisting of *A. sclerotiorum*, *A. persii*, *A. bridgeri*, and *A. sulphureus*, the three first species produce notoamides.

In most other clades in *Penicillium* and *Aspergillus* one, or rarely two, out of all species in a series produces bicyclo[2.2.2]-diazaoctane compounds (Table 10). Thus, metabolites containing the dioxopiperazine and prenylation are very widespread in *Aspergillus* and *Penicillium*, but the more complicated metabolites containing the bicyclo[2.2.2]diazaoctane ring system are present only in a few species. While it is only *P. paneum* that can produce marcfortines in *Penicillium* series *Roqueforti*, all species in the series can produce roquefortine C, another prenylated diketopiperazine (Trp + His). In the eight species in *Penicillium* series *Viridicatum* it is only *P. viridicatum* that can produce brevianamide A, but all eight species produce other prenylated dioxopiperazines:^{22,23} *P. neoehinulatum*, *P. aurantiogriseum*, and *P. freii* produce aurantiamine (Val + His), *P. viridicatum* produces the closely related viridamine (Val + His), *P. melanoconidium* produces roquefortines (Trp + His), and *P. cyclopium*, *P. polonicum*, and *P. tricolor* produce verrucofortines (Trp + Leu) and puberulines (Trp + Phe). In the five species in *Penicillium* series *Chrysogena* it is only *P. chrysogenum* that can produce chrysogenamide, but all species produce either prenylated dioxopiperazines or unprenylated dioxopiperazines: *P. chrysogenum*, *P. rubens*, and *P. flavigenum* produce roquefortine C (Trp + His), *P. dipodomyis* and *P. nalgiovense* produce dipodazin (Trp + Gly, unprenylated), and *P. chrysogenum* produces chrysogenazine (Trp + Ala, prenylated).⁶² The fact that prenylated dioxopiperazines are widespread in *Penicillium* and *Aspergillus*, and also occur outside these two genera, and the fact that the bicyclo[2.2.2]diazaoctane ring system-containing secondary metabolites are more restricted in their distribution suggest that these latter natural products are advanced molecules that appeared later during evolution than other dioxopiperazines.

■ CONCLUSION

Natural products containing the unique bicyclo[2.2.2]-diazaoctane ring system are intriguing from a synthetic, biosynthetic, evolutionary, and medicinal perspective. This unusual ring system is proposed to arise biosynthetically via an azadiene-mediated intramolecular hetero-Diels–Alder reaction that has been supported through biomimetic syntheses. An emerging body of evidence suggests that there are two distinct biogenetic families: (1) those constituted of a dioxopiperazine as an integral part of the bicyclo[2.2.2]diazaoctane ring system and (2) those comprised of a monooxopiperazine as an integral part of the bicyclo[2.2.2]diazaoctane ring system. Precursor incorporation experiments and recent genetic, bioinformatic, and biochemical analyses further suggest that

the dioxopiperazine-containing family members, such as stephacidins, arise by oxidative processing of C-2 reverse-prenylated indole dipeptides that culminate in azadiene formation and intramolecular Diels–Alder cyclization. Similarly, precursor incorporation experiments and recent analysis of the NRPS domains corresponding to the monooxopiperazine-containing family members indicate that natural products in the larger paraherquamide family arise by reductive processing of the initial indole dipeptides, to form a reduced azadiene species, which undergoes C-2 reverse prenylation and an intramolecular Diels–Alder cyclization. Whether or not the putative Diels–Alderase are none other than the oxidase and reductase domains of the NRPS modules that process the dipeptide intermediates at an early stage remains to be ascertained. At the heart of this central question is whether these redox functions catalyze the IMDA reaction through a pericyclic transition state, or whether the IMDA reaction is an “afterthought” or spontaneous construction immediately following the generation of the respective azadiene species that have been implicated. One of the most provocative findings in this context has been the discovery of the enantiomeric natural products stephacidin A, notoamide B, and versicolamide B, as discussed above, as well as the production of the diastereomeric and pseudoenantiomeric substances VM55599 and pre-paraherquamide. These intriguing stereochemical puzzles remain to be solved and are the subject of central focus in our laboratories.

The fungal origin of secondary metabolites containing the bicyclo[2.2.2] nucleus is largely limited to species within the *Aspergillus* and *Penicillium* genera. Moreover, secondary metabolites with the bicyclo[2.2.2] nucleus appear to be different from the more widespread prenylated dioxopiperazines in that they have evolved later and followed recent speciation in *Aspergillus* and *Penicillium*. Comparison of the malbrancheamide biosynthetic genes in *Malbranchea* with the parallel ones in *Penicillium* and *Aspergillus* may help clarify whether secondary metabolite gene clusters are acquired vertically or horizontally or if they have developed independently during evolution. It is tempting to ponder if the putative Diels–Alderase genes have evolved at a different rate than the amino acid cyclases and prenyl transferases in the producing organisms, a hypothesis that can be tested using bioinformatics combined with taxonomy and organic chemistry. The key step that culminates in the formation of the bicyclo[2.2.2]diazaoctane ring system is also the stage at which the structural complexity of the simple prenylated dipeptide progenitors are vastly increased. That Nature has apparently evolved two distinct mechanistic manifolds (reductive and oxidative) to build this ring system was an unexpected yet wondrous discovery that was only made possible through the synthesis of isotopically labeled biosynthetic intermediates, whole genome sequencing, and bioinformatic analyses of the respective gene clusters. The modern tools of whole genome sequencing, proteomics, bioinformatics, and functional expression of the biosynthetic genes are currently being deployed to gain a detailed understanding of the elusive mysteries of the biogenesis of these structurally complex secondary metabolites.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: rmw@lamar.colostate.edu.

Notes

The authors declare no competing financial interest.

REFERENCES

- (1) Li, S.; Anand, K.; Tran, H.; Yu, F.; Finefield, J. M.; Sunderhaus, J. D.; McAfoos, T. J.; Tsukamoto, S.; Williams, R. M.; Sherman, D. H. *Med. Chem. Commun.* **2012**, DOI: 10.1039/c2md20029e.
- (2) Sunderhaus, J. D.; Sherman, D. H.; Williams, R. M. *Isr. J. Chem.* **2011**, *51*, 442–452.
- (3) (a) Birch, A. J.; Wright, J. J. *J. Chem. Soc., Chem. Commun.* **1969**, 644–645. (b) Birch, A. J.; Wright, J. J. *Tetrahedron* **1970**, *26*, 2329–2344. (c) Birch, A. J.; Russel, R. A. *Tetrahedron* **1972**, *28*, 2999–3008. (d) Robbers, J. E.; Straus, J. W.; Tuite, J. J. *Nat. Prod.* **1975**, *38*, 355–356. (e) Wilson, B. J.; Yang, D. T. C.; Harris, T. M. *Appl. Microbiol.* **1973**, *26*, 633–635.
- (4) (a) Yamazaki, M.; Okuyama, E.; Kobayashi, M.; Inoue, H. *Tetrahedron Lett.* **1981**, *22*, 135–136. (b) Ondeyka, J. G.; Goegelman, R. T.; Schaeffer, J. M.; Kelemen, L.; Zitano, L. J. *Antibiot.* **1990**, *43*, 1375–1379. (c) Goegelman, R.; Ondeyka, J. U.S. Patent 4,873,247, 1987. (d) Lopez-Gresa, M. P.; Gonzalez, M. C.; Ciavatta, L.; Ayala, I.; Moya, P.; Primo, J. J. *Agric. Food Chem.* **2006**, *54*, 2921–2925. (e) Blanchflower, S. E.; Bank, R. M.; Everett, J. R.; Reading, C. J. *Antibiot.* **1993**, *46*, 1355–1363. (f) Banks, R. M.; Blanchflower, S. E.; Everett, J. R.; Manger, B. R.; Reading, C. J. *Antibiot.* **1997**, *50*, 840–846. (g) Blanchflower, S. E.; Banks, R. M.; Everett, J. R.; Manger, B. R.; Reading, C. J. *Antibiot.* **1991**, *44*, 492–497. (h) Liesch, J. M.; Wichmann, C. F. J. *Antibiot.* **1990**, *43*, 1380–1386. (i) Nielsen, K. F.; Sumarah, M. W.; Frisvad, J. C.; Miller, J. D. J. *Agric. Food Chem.* **2006**, *54*, 3756–3763. (j) Ding, Y.; Gruschow, S.; Greshock, T. J.; Finefield, J. M.; Sherman, D. H.; Williams, R. M. *J. Nat. Prod.* **2008**, *71*, 1574–1578. (k) Antia, B. S.; Aree, T.; Kasetrathat, C.; Wiyakrutta, S.; Ekpa, O. P.; Ekpe, U. J.; Mahindol, C.; Ruchirawat, S.; Kittakop, P. *Phytochemistry* **2011**, *72*, 816–820.
- (5) (a) Whyte, A. C.; Gloer, J. B.; Wicklow, D. T.; Dowd, P. F. *J. Nat. Prod.* **1996**, *59*, 1093–1095. (b) Qian-Cutrone, J.; Krampitz, K. D.; Shu, Y.-Z.; Chang, L.-P.; Lowe, S. E. U.S. Patent 6,291,461, 2001. (c) Qian-Cutrone, J.; Huang, S.; Shu, Y.-Z.; Vydas, D.; Fairchild, C.; Menendez, A.; Krampitz, K.; Dalterio, R.; Kloor, S. E.; Gao, Q. *J. Am. Chem. Soc.* **2002**, *124*, 14556–14557. (d) Fenical, W.; Jensen, P. R.; Cheng, X. C. U.S. Patent 6,066,635, 2000. (e) Sugie, Y.; Hirai, H.; Inagaki, T.; Ishiguro, M.; Kim, Y.-J.; Kojima, Y.; Sakakibara, T.; Sakemi, S.; Sugiura, A.; Suzuki, Y.; Brennan, L.; Duignan, J.; Huang, L. H.; Sutcliffe, J.; Kojima, N. J. *Antibiot.* **2001**, *54*, 911–916.
- (6) (a) Kato, H.; Yoshida, T.; Tokue, T.; Nojiri, Y.; Hirota, H.; Ohta, T.; Williams, R. M.; Tsukamoto, S. *Angew. Chem., Int. Ed.* **2007**, *46*, 2254–2256. (b) Tsukamoto, S.; Kato, H.; Greshock, T. J.; Hirota, H.; Ohta, T.; Williams, R. M. *J. Am. Chem. Soc.* **2009**, *131*, 3834–3835. (c) Tsukamoto, S.; Kato, H.; Samizo, M.; Nojiri, Y.; Onuki, H.; Hirota, H.; Ohta, T. *J. Nat. Prod.* **2008**, *71*, 2064–2067. (d) Tsukamoto, S.; Kawabata, T.; Kato, H.; Greshock, T. J.; Hirota, H.; Ohta, T.; Williams, R. M. *Org. Lett.* **2009**, *11*, 1297–1300. (e) Tsukamoto, S.; Umaoka, H.; Yoshikawa, K.; Ikeda, T.; Hirota, H. *J. Nat. Prod.* **2010**, *73*, 1438–1440. (f) Greshock, T. J.; Grubbs, A. W.; Jiao, P.; Wicklow, D. T.; Gloer, J. B.; Williams, R. M. *Angew. Chem., Int. Ed.* **2008**, *47*, 3573–3577. (g) Kito, K.; Ookura, R.; Kusumi, T.; Namikoshi, M.; Ooi, T. *Heterocycles* **2009**, *78*, 2101–2106.
- (7) (a) Hayashi, H.; Nishimoto, Y.; Nozaki, H. *Tetrahedron Lett.* **1997**, *38*, 5655–5658. (b) Hayashi, H.; Nishimoto, Y.; Akiyama, K.; Nozaki, H. *Biosci. Biotechnol. Biochem.* **2000**, *64*, 111–115. (c) Banks, R. M.; Blanchflower, S. E.; Everett, J. R.; Manger, B. R.; Reading, C. J. *Antibiot.* **1997**, *50*, 840–846.
- (8) (a) Polonsky, J.; Merrien, M.-A.; Prangé, T.; Pascard, C.; Moreau, S. J. *J. Chem. Soc., Chem. Commun.* **1980**, 601–602. (b) Prangé, T.; Billion, M.-A.; Vuilhorgien, M.; Pascard, C.; Polonsky, J. *Tetrahedron Lett.* **1981**, *22*, 1977–1980. (c) Capon, R. J.; Skene, C.; Stewart, M.; Ford, J.; O'Hair, R. A. J.; Williams, L.; Lacey, E.; Gill, J. H.; Heiland, K. *Org. Biomol. Chem.* **2003**, *1*, 1856–1862. (d) Nielsen, K. F.; Sumarah, M. W.; Frisvad, J. C.; Miller, J. D. J. *Agric. Food Chem.* **2006**, *54*, 3756–3763. (e) O'Brien, M.; Nielsen, K. F.; O'Kiely, P.; Forristal, P. D.; Fuller, H. T.; Frisvad, J. C. *J. Agric. Food Chem.* **2006**, *54*, 9268–9276. (f) Lee, B. H.; Clothier, M. F.; Kornis, G. I. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Elsevier Science: Oxford, U.K., 1998; Vol. 12, Chapter 6, pp 344–374.
- (9) Lin, Z.; Wen, J.; Zhu, T.; Fang, Y.; Gu, Q.; Zhu, W. *J. Antibiot.* **2008**, *61*, 81–85.
- (10) (a) Martínez-Luis, S.; Rodríguez, R.; Acevedo, L.; Gonzales, M. C.; Lira-Rocha, A.; Mata, R. *Tetrahedron* **2006**, *62*, 1817–1822. (b) Figueroa, M.; González, M. D. C.; Mata, R. *Nat. Prod. Res.* **2008**, *22*, 709–714. (c) Figueroa, M.; Gonzales, M.; Mata, R. *Planta Med.* **2008**, *74*, PB92. (d) Watts, K. R.; Loveridge, S. T.; Tenney, K.; Media, J.; Valeriotte, F. A.; Crews, P. *J. Org. Chem.* **2011**, *76*, 6201–6208. (e) Martínez-Luis, S.; González, M. C.; Ulloa, M.; Mata, R. *Phytochemistry* **2005**, *66*, 1012–1016. (f) Ding, Y.; Greshock, T. J.; Miller, K. A.; Sherman, D. H.; Williams, R. M. *Org. Lett.* **2008**, *10*, 4863–4866.
- (11) Stocking, E. M.; Sanz-Cervera, J. F.; Williams, R. M. *Angew. Chem., Int. Ed.* **1999**, *38*, 786–789.
- (12) Stocking, E. M.; Williams, R. M.; Sanz-Cervera, J. F. *J. Am. Chem. Soc.* **2000**, *122*, 9089–9098.
- (13) Williams, R. M.; Cox, R. J. *Acc. Chem. Res.* **2003**, *36*, 127–139.
- (14) Williams, R. M.; Sanz-Cervera, J. F.; Stocking, E. M. In *Topics in Current Chemistry, Vol. on Biosynthesis-Terpenes and Alkaloids*; Lepper, F.; Vederas, J. C., Eds.; Springer: Berlin, 2000; Vol. 209, pp 97–173.
- (15) Miller, K. A.; Williams, R. M. *Chem. Soc. Rev.* **2009**, *38*, 3160–3174.
- (16) (a) Porter, A. E. A.; Sammes, P. G. *Chem. Commun.* **1970**, 1103. (b) Williams, R. M. *Chem. Pharm. Bull.* **2002**, *50*, 711–740.
- (17) (a) Williams, R. M.; Sanz-Cervera, J. F.; Marco, J. A.; Halligan, K. *J. Am. Chem. Soc.* **1998**, *120*, 1090–1091. (b) Williams, R. M.; Sanz-Cervera, J. F.; Sancenón, F.; Marco, J. A.; Halligan, K. *Bioorg. Med. Chem.* **1998**, *6*, 1233–1241. (c) Greshock, T. J.; Grubbs, A. W.; Tsukamoto, S.; Williams, R. M. *Angew. Chem., Int. Ed.* **2007**, *46*, 2262–2265. (d) Greshock, T. J.; Williams, R. M. *Org. Lett.* **2007**, *9*, 4255–4258. (e) Greshock, T. J.; Grubbs, A. W.; Williams, R. M. *Tetrahedron* **2007**, *63*, 6124–6130. (f) Miller, K. A.; Tsukamoto, S.; Williams, R. M. *Nat. Chem.* **2009**, *1*, 63–68. (g) Domingo, L. R.; Zaragoza, R. J.; Williams, R. M. *J. Org. Chem.* **2003**, *68*, 2895–2902. (h) Stocking, E. M.; Sanz-Cervera, J. F.; Williams, R. M. *J. Am. Chem. Soc.* **2000**, *122*, 1675–1683.
- (18) Raper, K. B.; Thom, C. *Manual of the Penicillia*; Williams and Wilkins: Baltimore, 1949.
- (19) Raper, K. B.; Fennell, D. I. *The Genus Aspergillus*; William and Wilkins: Baltimore, 1965.
- (20) Miller, K. A.; Figueroa, M.; Valente, M. W. N.; Greshock, T. J.; Mata, R.; Williams, R. M. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 6479–6481.
- (21) Miller, K. A.; Welch, T. R.; Greshock, T. J.; Ding, Y.; Sherman, D. H.; Williams, R. M. *J. Org. Chem.* **2008**, *73*, 3116–3119.
- (22) Frisvad, J. C.; Smedsgaard, J.; Larsen, T. O.; Samson, R. A. *Stud. Mycol.* **2004**, *49*, 201–241.
- (23) Frisvad, J. C.; Samson, R. A. *Stud. Mycol.* **2004**, *49*, 1–173.
- (24) Houbraken, J.; Frisvad, J. C.; Samson, R. A. *IMA Fungus* **2011**, *2*, 87–95.
- (25) (a) Paterson, R. R. M.; Simmonds, M. S. J.; Blaney, W. M. *J. Invertebr. Pathol.* **1987**, *50*, 124–133. (b) Paterson, R. R. M.; Simmonds, M. S. J.; Kimmelmeyer, C.; Blaney, W. M. *Mycol. Res.* **1990**, *94*, 538–542.
- (26) Frisvad, J. C.; Filtenborg, O. *Mycologia* **1989**, *81*, 836–851.
- (27) (a) Steyn, P. S. *Tetrahedron* **1971**, *36*, 3331–3334. (b) Steyn, P. S. *Tetrahedron* **1973**, *29*, 107–120. (c) Steyn, P. S.; Vlegaar, R. *Phytochemistry* **1976**, *15*, 355–356. (d) Scott, P. M.; Kennedy, B. P. C.; Harwig, J.; Chen, Y.-K. *Appl. Microbiol.* **1974**, *28*, 892–894. (e) Arai, K.; Miyajima, H.; Mushiroda, T.; Yamamoto, Y. *Chem. Pharm. Bull.* **1989**, *37*, 3229–3235.
- (28) Samson, R. A.; Varga, J.; Meijer, M.; Frisvad, J. C. *Stud. Mycol.* **2011**, *69*, 81–97.
- (29) Frisvad, J. C.; Rank, C.; Nielsen, K. F.; Larsen, T. O. *Med. Mycol.* **2009**, *47*, S53–S71.
- (30) Zhang, D.; Novindri, D.; Nursid, M.; Yang, X.; Son, B. W. *Nat. Prod. Sci.* **2007**, *13*, 251–254.

- (31) Baldas, J.; Birch, A. J.; Russell, R. A. *J. Chem. Soc., Perkin Trans. 1* **1974**, 50–52.
- (32) Sanz-Cervera, J. F.; Glinka, T.; Williams, R. M. *Tetrahedron* **1993**, 49, 8471–8482.
- (33) Boysen, M.; Skouboe, P.; Frisvad, J. C.; Rossen, L. *Microbiology-UK* **1996**, 142, 541–549.
- (34) (a) Kuo, M. S.; Wiley, V. H.; Cialdella, J. I.; Yurek, D. A.; Whaley, H. A.; Marshall, V. P. *J. Antibiot.* **1996**, 49, 1006–1013. (b) Kuo, M. S.; Yurek, D. A.; Mizesak, S. A.; Cialdella, J. I.; Baczynskyj, L.; Marshall, V. P. *J. Am. Chem. Soc.* **1999**, 121, 1763–1767.
- (35) (a) Filtenborg, O.; Frisvad, J. C. In *Modern Concepts in Penicillium and Aspergillus Classification*; Samson, R. A.; Pitt, J. I., Eds.; Plenum Press: New York, 1990; pp 159–170. (b) Tuthill, D. E.; Frisvad, J. C.; Christensen, M. *Mycologia* **2001**, 93, 298–308.
- (36) (a) Frisvad, J. C.; Samson, R. A.; Stolk, A. C. *Persoonia* **1990**, 14, 193–202. (b) Frisvad, J. C.; Samson, R. A.; Stolk, A. C. *Persoonia* **1990**, 14, 209–232.
- (37) Pitt, J. I. *The Genus Penicillium and Its Teleomorphic States Eupenicillium and Talaromyces*; Academic Press: New York, 1992.
- (38) (a) Byung, H. L.; Clothier, M. F.; Dutton, F. E.; Nelson, S. J.; Johnson, S. S.; Thompson, D. P.; Geary, T. G.; Whaley, H. D.; Haber, C. L.; Marshall, V. P.; Kornis, G. I.; McNally, P. L.; Cialdella, J. I.; Martin, D. G.; Bowman, J. W.; Baker, C. A.; Coscarelli, E. M.; Alexander-Bowman, S. J.; Davis, J. P.; Zinser, E. W.; Wiley, V.; Lipton, M. F.; Mauragis, M. A. *Curr. Top. Med. Chem.* **2002**, 2, 779–793. (b) Lee, B. H.; Clothier, M. F.; Johnson, S. S. *Bioorg. Med. Chem. Lett.* **2001**, 11, 553–554.
- (39) Ostlind, D. A.; Mickle, W. G.; Ewanciw, D. V.; Andriuli, F. J.; Campbell, W. C.; Hernandez, S.; Mochales, S.; Munguira, E. *Res. Vet. Sci.* **1990**, 48, 260–261.
- (40) Zinser, E. W.; Wolfe, M. L.; Alexander-Bowman, S. J.; Thomas, E. M.; Davis, J. P.; Groppi, V. E.; Lee, B. H.; Thompson, D. P.; Geary, T. G. *J. Vet. Pharmacol. Ther.* **2002**, 25, 241–250.
- (41) Shoop, W. L.; Haines, H. W.; Eary, C. H.; Michael, B. F. *Am. J. Vet. Res.* **1992**, 53, 2032–2034.
- (42) (a) Kaminsky, R.; Bapst, B.; Stein, P. A.; Strehlau, G. A.; Allan, B. A.; Hosking, B. C.; Rolfe, P. F.; Sager, H. *Parasitol. Res.* **2011**, 109, 19–23. (b) Love, S. *AFBM J.* **2010**, 7, 45–52. (c) Little, P. R.; Hodges, A.; Watson, T. G.; Seed, J. A.; Maeder, S. J. *N. Z. Vet. J.* **2010**, 58, 121–129.
- (43) (a) Stocking, E. M.; Sanz-Cervera, J. F.; Unkefer, C. J.; Williams, R. M. *Tetrahedron* **2001**, 57, 5303–5320. (b) Kellenberger, J. L. The Stereochemical Course of the C-methylation Steps in the Biosynthesis of Bottromycin. Ph.D. Thesis, ETH Zurich, Zurich, Switzerland, 1997. (c) Stocking, E. M.; Sanz-Cervera, J. F.; Williams, R. M.; Unkefer, C. J. *J. Am. Chem. Soc.* **1996**, 118, 7008–7009. (d) Stocking, E. M.; Martinez, R. A.; Silks, L. A.; Sanz-Cervera, J. F.; Williams, R. M. *J. Am. Chem. Soc.* **2001**, 123, 3391–3392.
- (44) Stocking, E. M.; Sanz-Cervera, J. F.; Williams, R. M. *Angew. Chem., Int. Ed.* **2001**, 40, 1296–1298.
- (45) Noonim, P.; Mahakarnchanakul, W.; Varga, J.; Frisvad, J. C.; Samson, R. A. *Int. J. Syst. Evol. Microbiol.* **2008**, 58, 1727–1734.
- (46) (a) Pareniková, L.; Skouboe, P.; Frisvad, J. C.; Samson, R. A.; Rossen, L.; Hoor Suykerbuyk, M.; Visser, J. *Appl. Environ. Microbiol.* **2001**, 67, 521–527. (b) Samson, R. A.; Houbraken, J. A. M. P.; Kuijpers, A. F. A.; Frank, J. M.; Frisvad, J. C. *Stud. Mycol.* **2004**, 50, 45–61. (c) Samson, R. A.; Noonim, P.; Meijer, M.; Houbraken, J.; Frisvad, J. C.; Varga, J. *Stud. Mycol.* **2007**, 59, 129–145. (d) Perrone, G.; Varga, J.; Susca, A.; Frisvad, J. C.; Stea, G.; Kocsubé, S.; Tóth, B.; Kozakiewicz, Z. *Int. J. Syst. Evol. Microbiol.* **2008**, 58, 1032–1039. (e) Sørensen, A.; Lübeck, P. S.; Lübeck, M.; Nielsen, K. F.; Ahring, B. K.; Teller, P. J.; Frisvad, J. C. *Int. J. Syst. Evol. Microbiol.* **2011**, 61, 3077–3083. (f) Varga, J.; Frisvad, J. C.; Kocsubé, S.; Brancovics, B.; Tóth, B.; Szigeti, G.; Samson, R. A. *Stud. Mycol.* **2011**, 69, 1–17.
- (47) Gray, C. R.; Sanz-Cervera, J. F.; Silks, L. A.; Williams, R. M. *J. Am. Chem. Soc.* **2003**, 125, 14692–14693.
- (48) The structures of the norgeamides were published via the Internet detailing the research performed by the Hans-Knöll Institute. www.hki-jena.de.
- (49) McAfoos, T. J.; Li, S.; Tsukamoto, S.; Sherman, D. H.; Williams, R. M. *Heterocycles* **2010**, 82, 461–472.
- (50) Frisvad, J. C.; Frank, J. M.; Houbraken, J. A. M. P.; Kuijpers, A. F. A.; Samson, R. A. *Stud. Mycol.* **2004**, 50, 23–43.
- (51) Grubbs, A. W.; Artman, G. D. III; Williams, R. M. *Tetrahedron Lett.* **2005**, 46, 9013–9016.
- (52) (a) Myers, A. G.; Herzon, S. B. *J. Am. Chem. Soc.* **2003**, 125, 12080–12081. (b) Wulff, J. E.; Herzon, S. B.; Siegrist, R.; Myers, A. G. *J. Am. Chem. Soc.* **2007**, 129, 4898–4899.
- (53) Ding, Y.; de Wet, J. R.; Cavalcoli, J.; Li, S.; Greshock, T. J.; Miller, K. A.; Finefield, J. M.; Sunderhaus, J. D.; McAfoos, T. J.; Tsukamoto, S.; Williams, R. M.; Sherman, D. H. *J. Am. Chem. Soc.* **2010**, 132, 12733–12740.
- (54) Finefield, J. M.; Kato, H.; Greshock, T. J.; Sherman, D. H.; Tsukamoto, S.; Williams, R. M. *Org. Lett.* **2011**, 3802–3805.
- (55) (a) Tsuda, M.; Kasai, Y.; Komatsu, K.; Sone, T.; Tanaka, M.; Mikami, Y.; Kobayashi, J. *Org. Lett.* **2004**, 6, 3087–3089. (b) Mugishima, T.; Tsuda, M.; Kasai, Y.; Ishiyama, H.; Fukushi, E.; Kawabata, J.; Watanabe, M.; Akao, K.; Kobayashi, J. *J. Org. Chem.* **2005**, 70, 9430–9435.
- (56) (a) Kushida, N.; Watanabe, N.; Okuda, T.; Yokoyama, F.; Gyobu, Y.; Yaguchi, T. *J. Antibiot.* **2007**, 60, 667–673. (b) Kushida, N.; Watanabe, N.; Yaguchi, T.; Yokoyama, F.; Tsujiuchi, G.; Okuda, T. U.S. Patent 7,501,431, 2009.
- (57) (a) Houbraken, J.; Frisvad, J. C.; Samson, R. A. *Fungal Diversity* **2010**, 44, 117–133. (b) Houbraken, J.; Frisvad, J. C.; Samson, R. A. *Stud. Mycol.* **2011**, 70, 53–138.
- (58) (a) Chu, Y.-S.; Niu, X.-M.; Wang, Y.-L.; Guo, J.-P.; Pan, W.-Z.; Huang, X.-W.; Zhang, K.-Q. *Org. Lett.* **2010**, 12, 4356–4357. (b) Guo, J.-P.; Tan, J.-L.; Wang, Y.-L.; Wu, H.-Y.; Zhang, C.-P.; Niu, X.-M.; Pan, W.-Z.; Huang, X.-W.; Zhang, K.-Q. *J. Nat. Prod.* **2011**, 74, 2278–2281.
- (59) Davies, J. *Mol. Microbiol.* **1990**, 4, 1227–1232.
- (60) Peterson, S. W. In *Integration of Modern Taxonomic Methods for Penicillium and Aspergillus Classification*; Samson, R. A.; Pitt, J. I., Eds.; Harwood: Amsterdam, 2000; pp 323–355.
- (61) (a) Peterson, S. W. In *Integration of Modern Taxonomic Methods for Penicillium and Aspergillus Classification*; Samson, R. A.; Pitt, J. I., Eds.; Harwood: Amsterdam, 2000; pp 163–178. (b) Houbraken, J.; Samson, R. A. *Stud. Mycol.* **2011**, 70, 1–51. (c) Samson, R. A.; Yilmaz, N.; Houbraken, J.; Spierenburg, H.; Seifert, K. A.; Peterson, S. W.; Varga, J.; Frisvad, J. C. *Stud. Mycol.* **2011**, 70, 159–183. (d) Houbraken, J.; Spierenburg, H.; Frisvad, J. C. *Antonie van Leeuwenhoek* **2012**, 101, 403–421.
- (62) Naik, C. G.; Devi, P.; Rodrigues, E. U.S. Patent 2005143392, 2005.
- (63) Nielsen, K. F.; Månsson, M.; Rank, C.; Frisvad, J. C.; Larsen, T. O. *J. Nat. Prod.* **2011**, 74, 2338–2348.